

MONOAMINE REUPTAKE INHIBITORS FOR
TREATMENT OF CNS DISORDERS

The present application is a continuation-in-part of U.S. Serial No. 09/529,207, which was filed on April 7, 2000 as the U.S. national phase of World Patent Application
5 PCT/IB00/00108, which was filed in the International Bureau of the PCT on February 2, 2000. The present application claims priority from both the foregoing U.S. and World patent applications, as well as from U.S. Provisional Application Serial No. 60/121,313.

Background of the Invention

Serotonin Selective Reuptake Inhibitors (SSRIs) currently provide efficacy in the
10 treatment of major depressive disorder (MDD) and are generally perceived by psychiatrists and primary care physicians as effective, well-tolerated and easily administered. However, they are associated with undesirable features, such as high incidence of sexual dysfunction, delayed onset of action and a level of non-responsiveness estimated to be as high as 30% (see M. J. Gitlin, Journal of Clinical Psychiatry, **1994**, 55, 406-413 and R. T. Segraves, Journal
15 of Clinical Psychiatry, **1992**, 10(2), 4-10). Preclinical and clinical evidence has indicated that the sexual dysfunction associated with SSRI therapy can be reduced through the use of dopamine reuptake inhibitors (DRIs), such as bupropion (see A. K. Ashton, Journal of Clinical Psychiatry, **1998**, 59(3), 112-115). Furthermore, the combination of SRI and DRI may hasten the onset of action as well as offering relief to refractory patients, possibly through a
20 synergistic mechanism (see R. D. Marshall et al, Journal of Psychopharmacology, **1995**, 9(3), 284-286).

This invention relates to novel biaryl ether derivatives that exhibit activity as
monoamine (e.g., dopamine, serotonin) reuptake inhibitors, to pharmaceutical compositions containing such compounds and to methods of using such compounds to treat central
25 nervous system (CNS) and other disorders.

United States Patent 4,018,830, published on April 19, 1997, refers to phenylthioaralkylamines and 2-phenylthiobenzylamines which are active as antiarrhythmics.

PCT publication 97/17325, published on May 15, 1997, refers to derivatives of N,N-dimethyl-2-(arylthio)benzylamine which selectively influence serotonin transport in the central
30 nervous system and are useful as antidepressants.

United States Patent 5,190,965, published on March 2, 1993, and United States Patent 5,430,063, issued July 4, 1995, refer to phenoxyphenyl derivatives which have utility in the treatment of depression.

United States Patent 4,161,529, published on July 17, 1979, refers to pyrrolidine
35 derivative which possess anticholesteremic and hypolipemic activity.

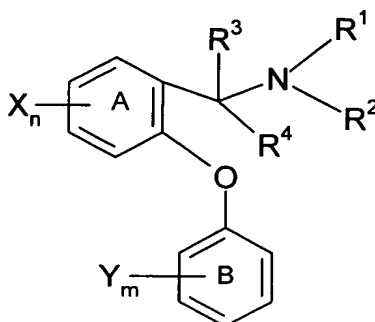
Summary of the Invention

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The present invention relates to compounds of the formula



wherein phenyl ring A and phenyl ring B can each, independently, be replaced by a naphthyl group, and wherein when phenyl ring A is replaced by a naphthyl group, the ethereal oxygen of structure I and the carbon to which R³, R⁴ and NR¹R² are attached, are attached to adjacent ring carbon atoms of the naphthyl group and neither of said adjacent ring carbon atoms is also adjacent to a fused ring carbon atom of said naphthyl group;

n and m are, selected, independently, from one, two and three;

R¹ and R² are selected, independently, from hydrogen, (C₁-C₄)alkyl, (C₂-C₄)alkenyl, and (C₂-C₄)alkynyl, or R¹ and R², together with the nitrogen to which they are attached, form a four to eight membered saturated ring containing one or two heteroatoms, including the nitrogen to which R¹ and R² are attached, wherein the second heteroatom, when present, is selected from oxygen, nitrogen and sulfur, and wherein said ring may optionally be substituted at available binding sites with from one to three substituents selected, independently, from hydroxy and (C₁-C₆)alkyl;

R³ and R⁴ are selected, independently, from hydrogen and (C₁-C₄) alkyl optionally substituted with from one to three fluorine atoms, or R³ and R⁴, together with the carbon to which they are attached, form a four to eight membered saturated carbocyclic ring, and wherein said ring may optionally be substituted at available binding sites with from one to three substituents selected, independently, from hydroxy and (C₁-C₆)alkyl;

or R² and R³, together with the nitrogen to which R² is attached and the carbon to which R³ is attached, form a four to eight membered saturated ring containing one or two heteroatoms, including the nitrogen to which R² is attached, wherein the second heteroatom, when present, is selected from oxygen, nitrogen and sulfur, and wherein said ring may optionally be substituted at available binding sites with from one to three substituents selected, independently, from hydroxy and (C₁-C₆)alkyl;

each X is selected, independently, from hydrogen, halo (i.e., chloro, fluoro, bromo or iodo), (C₁-C₄)alkyl optionally substituted with from one to three fluorine atoms, (C₁-C₄)alkoxy optionally substituted with from one to three fluorine atoms, cyano, nitro, amino, (C₁-

each Y is selected, independently, from hydrogen, (C₁-C₆)alkyl and halo;

and the pharmaceutically acceptable salts thereof.

The present invention also relates to a pharmaceutical composition for treating a disorder or condition that can be treated by inhibiting the reuptake of serotonin, dopamine or norepinephrine in a mammal, preferably a human, comprising an amount of a compound of formula I, or a pharmaceutically acceptable salt thereof, that is effective in treating such disorder or condition and a pharmaceutically acceptable carrier. Examples of such disorders and conditions are those enumerated in the preceding paragraph.

The present invention also relates to a method for treating a disorder or condition selected from hypertension, depression (e.g., depression in cancer patients, depression in Parkinson's patients, postmyocardial infarction depression, subsyndromal symptomatic depression, depression in infertile women, pediatric depression, major depression, single episode depression, recurrent depression, child abuse induced depression, and post partum depression), generalized anxiety disorder, phobias (e.g., agoraphobia, social phobia and simple phobias), posttraumatic stress syndrome, avoidant personality disorder, premature ejaculation, eating disorders (e.g., anorexia nervosa and bulimia nervosa), obesity, chemical dependencies (e.g., addictions to alcohol, cocaine, heroin, phenobarbital, nicotine and benzodiazepines), cluster headache, migraine, pain, Alzheimer's disease, obsessive-compulsive disorder, panic disorder, memory disorders (e.g., dementia, amnesic disorders, and age-related cognitive decline (ARCD)), Parkinson's diseases (e.g., dementia in Parkinson's disease, neuroleptic-induced parkinsonism and tardive dyskinesias), endocrine disorders (e.g., hyperprolactinaemia), vasospasm (particularly in the cerebral vasculature), cerebellar ataxia, gastrointestinal tract disorders (involving changes in motility and secretion), negative symptoms of schizophrenia, premenstrual syndrome, fibromyalgia syndrome, stress incontinence, Tourette's syndrome, trichotillomania, kleptomania, male impotence, attention deficit hyperactivity disorder (ADHD), chronic paroxysmal hemicrania and headache (associated with vascular disorders) in a mammal, preferably a human, comprising administering to a mammal in need of such treatment an amount of a compound of the formula I, or a pharmaceutically acceptable salt thereof, that is effective in treating such disorder or condition.

The present invention also relates to a method for treating a disorder or condition that can be treated by inhibiting the reuptake of serotonin, dopamine or norepinephrine in a mammal, preferably a human, comprising administering to a mammal in need of such treatment an amount of a compound of the formula I, or a pharmaceutically acceptable salt thereof, that is effective in treating such disorder or condition.

The present invention also relates to a pharmaceutical composition for treating a disorder or condition selected from hypertension, depression (e.g., depression in cancer patients, depression in Parkinson's patients, postmyocardial infarction depression, subsyndromal symptomatic depression, depression in infertile women, pediatric depression, major depression, single episode depression, recurrent depression, child abuse induced depression, and post partum depression), generalized anxiety disorder, phobias (e.g., agoraphobia, social phobia and simple phobias), posttraumatic stress syndrome, avoidant personality disorder, premature ejaculation, eating disorders (e.g., anorexia nervosa and bulimia nervosa), obesity, chemical dependencies (e.g., addictions to alcohol, cocaine, heroin,

The present invention also relates to a method for treating a disorder or condition selected from hypertension, depression (e.g., depression in cancer patients, depression in Parkinson's patients, postmyocardial infarction depression, subsyndromal symptomatic depression, depression in infertile women, pediatric depression, major depression, single episode depression, recurrent depression, child abuse induced depression, and post partum depression), generalized anxiety disorder, phobias (e.g., agoraphobia, social phobia and simple phobias), posttraumatic stress syndrome, avoidant personality disorder, sexual dysfunction (e.g., premature ejaculation), eating disorders (e.g., anorexia nervosa and bulimia nervosa), obesity, chemical dependencies (e.g., addictions to alcohol, cocaine, heroin, phenobarbital, nicotine and benzodiazepines), cluster headache, migraine, pain, Alzheimer's disease, obsessive-compulsive disorder, panic disorder, memory disorders (e.g., dementia, amnesic disorders, and age-related cognitive decline (ARCD)), Parkinson's diseases (e.g., dementia in Parkinson's disease, neuroleptic-induced parkinsonism and tardive dyskinesias), endocrine disorders (e.g., hyperprolactinaemia), vasospasm (particularly in the cerebral vasculature), cerebellar ataxia, gastrointestinal tract disorders (involving changes in motility and secretion), negative symptoms of schizophrenia, premenstrual syndrome, fibromyalgia syndrome, stress incontinence, Tourette's syndrome, trichotillomania, kleptomania, male impotence, attention deficit hyperactivity disorder (ADHD), chronic paroxysmal hemicrania and

headache (associated with vascular disorders) in a mammal, preferably a human, comprising administering to a mammal requiring such treatment a serotonin, dopamine or norepinephrine reuptake inhibiting effective amount of a compound of the formula I, or a pharmaceutically acceptable salt thereof.

5 The present invention also relates to a method for treating a disorder or condition that can be treated by inhibiting the reuptake of serotonin, norepinephrine or dopamine in a mammal, preferably a human, comprising administering to a mammal requiring such treatment a serotonin, dopamine or norepinephrine reuptake inhibiting effective amount of a compound of the formula I, or a pharmaceutically acceptable salt thereof.

10 The present invention relates to a pharmaceutical composition for treating a condition or disorder that can be treated by inhibiting the reuptake of serotonin, dopamine or norepinephrine in a mammal, preferably a human, comprising:

- a) a pharmaceutically acceptable carrier;
- b) a compound of the formula I or a pharmaceutically acceptable salt thereof;

15 and

c) an NK-1 receptor antagonist or a 5HT_{1D} receptor antagonist, or a pharmaceutically acceptable salt thereof;

 wherein the amount of the active compounds (i.e., the compound of formula I and the NK-1 receptor antagonist or 5HT_{1D} receptor antagonist) are such that the combination is effective in treating such disorder or condition.

20 The present invention also relates to a method for treating a disorder or condition that can be treated by inhibiting the reuptake of serotonin, dopamine or norepinephrine in a mammal, preferably a human, comprising administering to a mammal requiring such treatment:

25 a) a compound of the formula I, defined above, or a pharmaceutically acceptable salt thereof; and

b) an NK-1 receptor antagonist or a 5HT_{1D} receptor antagonist, or a pharmaceutically acceptable salt thereof;

30 wherein the amounts of the active compounds (i.e., the compound of formula I and the NK-1 receptor antagonist or 5HT_{1D} receptor antagonist) are such that the combination is effective in treating such disorder or condition.

 This invention also relates to the pharmaceutically acceptable acid addition salts of the compounds of formula I. Examples of pharmaceutically acceptable acid addition salts of the compounds of formula I are the salts of hydrochloric acid, p-toluenesulfonic acid, fumaric acid, citric acid, succinic acid, salicylic acid, oxalic acid, hydrobromic acid, phosphoric acid,

35

Unless otherwise indicated, the term "halo", as used herein, includes fluoro, chloro, bromo and iodo.

The term “treatment”, as used herein, refers to reversing, alleviating, inhibiting the progress of, or preventing the disorder or condition to which such term applies, or one or more symptoms of such condition or disorder. The term “treatment”, as used herein, refers to the act of treating, as “treating” is defined immediately above.

The present invention also relates to all radiolabelled forms of the compounds of the formula I. Preferred radiolabelled compounds of formula I are those wherein the radiolabels are selected from as ^3H , ^{11}C , ^{14}C , ^{18}F , ^{123}I and ^{125}I . Such radiolabelled compounds are useful as research and diagnostic tools in metabolism pharmacokinetics studies and in binding assays in both animals and man.

Preferred embodiments of this invention include the following compounds of the formula I and their pharmaceutically acceptable salts:

- 30 [2-(3,4-Dichlorophenoxy)-5-fluorobenzyl]-dimethylamine;
[2-(3,4-Dichlorophenoxy)-5-fluorobenzyl]-methylamine;
[2-(3,4-Dichlorophenoxy)-5-trifluoromethylbenzyl]-dimethylamine;
N-[4-(3,4-Dichlorophenoxy)-3-dimethylaminomethylphenyl]-acetamide;
{1-[2-(3,4-Dichlorophenoxy)phenyl]-ethyl}-dimethylamine;
35 [2-(3,4-Dichlorophenoxy)-4-trifluoromethylbenzyl]-dimethylamine;
[2-(3,4-Dichlorophenoxy)-4-trifluoromethylbenzyl]-methylamine;

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- [4-Chloro-2-(3,4-dichlorophenoxy)-benzyl]-methylanine;
{1-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-ethyl}-methylanine;
{1-[2-(3,4-Dichlorophenoxy)phenyl]-ethyl}-methylanine;
{1-[2-(4-Chlorophenoxy)phenyl]ethyl}-methylanine;
5 [2-(3,4-Dichlorophenoxy)-5-methoxybenzyl]-methylanine;
[2-(4-Chlorophenoxy)-5-fluorobenzyl]-methylanine; and
{1-[2-(4-Chlorophenoxy)-5-fluorophenyl]-ethyl}-methylanine.
[2-(3,4-Dichlorophenoxy)-5-methylbenzyl]-dimethylanine;
[4-Bromo-2-(3,4-dichlorophenoxy)-benzyl]-methylanine;
10 [5-Bromo-2-(3,4-dichlorophenoxy)-benzyl]-methylanine;
[2-(3,4-Dichlorophenoxy)-4,5-dimethoxybenzyl]-methylanine;
[2-(3,4-Dichlorophenoxy)-4-methoxybenzyl]-dimethylanine;
4-(3,4-Dichlorophenoxy)-3-methylaminomethyl-benzonitrile;
[2-(3,4-Dichlorophenoxy)-4,5-dimethylbenzyl]-methylanine;
15 3-(3,4-Dichlorophenoxy)-4-methylaminomethyl-benzonitrile;
(+)-{1-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-ethyl}-methylanine;
(-)-{1-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-ethyl}-methylanine;
[2-(3,4-Dichlorophenoxy)-5-trifluoromethyl-benzyl]-methylanine;
[2-(3,4-Dichlorophenoxy)-4-methoxybenzyl]-methylanine;
20 [2-(4-Chloro-3-fluorophenoxy)-5-fluorobenzyl]-methylanine;
[2-(3-Chloro-4-fluorophenoxy)-5-fluorobenzyl]-methylanine;
(+/-)-2-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-pyrrolidine;
(-)-2-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-pyrrolidine;
(+)-2-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-pyrrolidine; and
25 2-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-N-methylpyrrolidine.

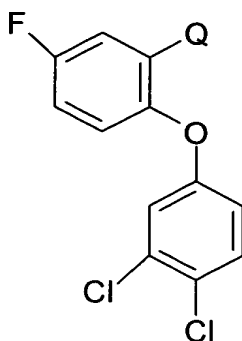
Other embodiments of this invention include the following compounds and their pharmaceutically acceptable salts:

- {1-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-1-methylethyl}-methylanine;
{1-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-1-methylethyl}-dimethylanine;
30 [4-Chloro-2-(4-chlorophenoxy)-5-fluorobenzyl]-methylanine;
[2-(3,4-Dichlorophenoxy)-5-fluoro-4-methoxybenzyl]-methylanine;
[4-(3,4-Dichlorophenoxy)-3-(dimethylaminomethyl)-phenyl]-dimethylanine
[5-Fluoro-2-(4-fluoro-3-methoxyphenoxy)-benzyl]-dimethylanine;
[2-(4-Chlorophenoxy)-5-isopropylbenzyl]-methylanine;
35 {1-[2-(4-Chlorophenoxy)-5-trifluoromethylphenyl]-ethyl}-methylanine;
[2-(4-Chlorophenoxy)-4,5-dimethylbenzyl]-methylanine;

- {1-[5-Chloro-2(3,4-dichlorophenoxy)phenyl]-propyl}-methylamine;
 [2-(3,4-Dichlorophenoxy)-5-methylsulfanyl-benzyl]-methylamine;
 {1-[2-(3,4-Dichlorophenoxy)-5-methylsulfanyl-phenyl]-ethyl}-methylamine;
 {1-[2-(3,4-Dichloro-phenoxy)-5-methylsulfanyl-phenyl]-1-methylethyl}-methylamine;
 5 [2-(3,4-Dichlorophenoxy)-5-methylsulfanyl-benzyl]-dimethylamine;
 [2-(3,4-Dichlorophenoxy)-5-methanesulfinyl-benzyl]-dimethylamine ;
 [2-(3,4-Dichlorophenoxy)-5-methanesulfinyl-benzyl]-methylamine;
 [2-(3,4-Dichlorophenoxy)-5-methanesulfonyl-benzyl]-methylamine;
 [2-(3,4-Dichlorophenoxy)-5-methanesulfonyl-benzyl]-dimethylamine;
 10 [2-(3,4-Dichlorophenoxy)-5-(propane-2-sulfonyl)-benzyl]-methylamine;
 2-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-piperidine;
 2-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-1-methyl-piperidine;
 3-[2-(3,4-Dichloro-phenoxy)-5-fluorophenyl]-4-methyl-morpholine;
 2-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-1,2-dimethyl-piperidine;
 15 {1-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-cyclopropyl}-dimethylamine;
 2-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-1,5-dimethyl-pyrrolidine;
 3-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-4-methyl-thiomorpholine;
 {1-[2-(3,4-Dichlorophenoxy)-5-fluorophenyl]-cyclopentyl}-methylamine;
 {1-[2-(3,4-Dichlorophenoxy)-5-(propane-2-sulfonyl)-phenyl]-ethyl}-methylamine; and
 20 [4-Chloro-2-(3,4-dichlorophenoxy)-5-methanesulfonyl-benzyl]-methylamine.

Other embodiments of this invention relate to the compound of the formula I wherein m is zero, n is one, R³ and R⁴ are hydrogen, X is chloro, bromo, iodo or methyl, R¹ is hydrogen and R² is methyl.

This invention also relates to compounds of the formula



XVIII

25 wherein Q is -C(=O)H, cyano, -C(=O)OH or -C(=O)NR¹R² wherein R¹ and R² are selected, independently, from hydrogen and (C₁-C₄)alkyl, or R¹ and R², together with the nitrogen to which they are attached, form a four to eight membered saturated ring containing one or two heteroatoms, including the nitrogen to which R¹ and R² are attached, wherein the second

heteroatom, when present, is selected from oxygen, nitrogen and sulfur. These compounds are useful as intermediates in the synthesis of certain compounds of the formula I.

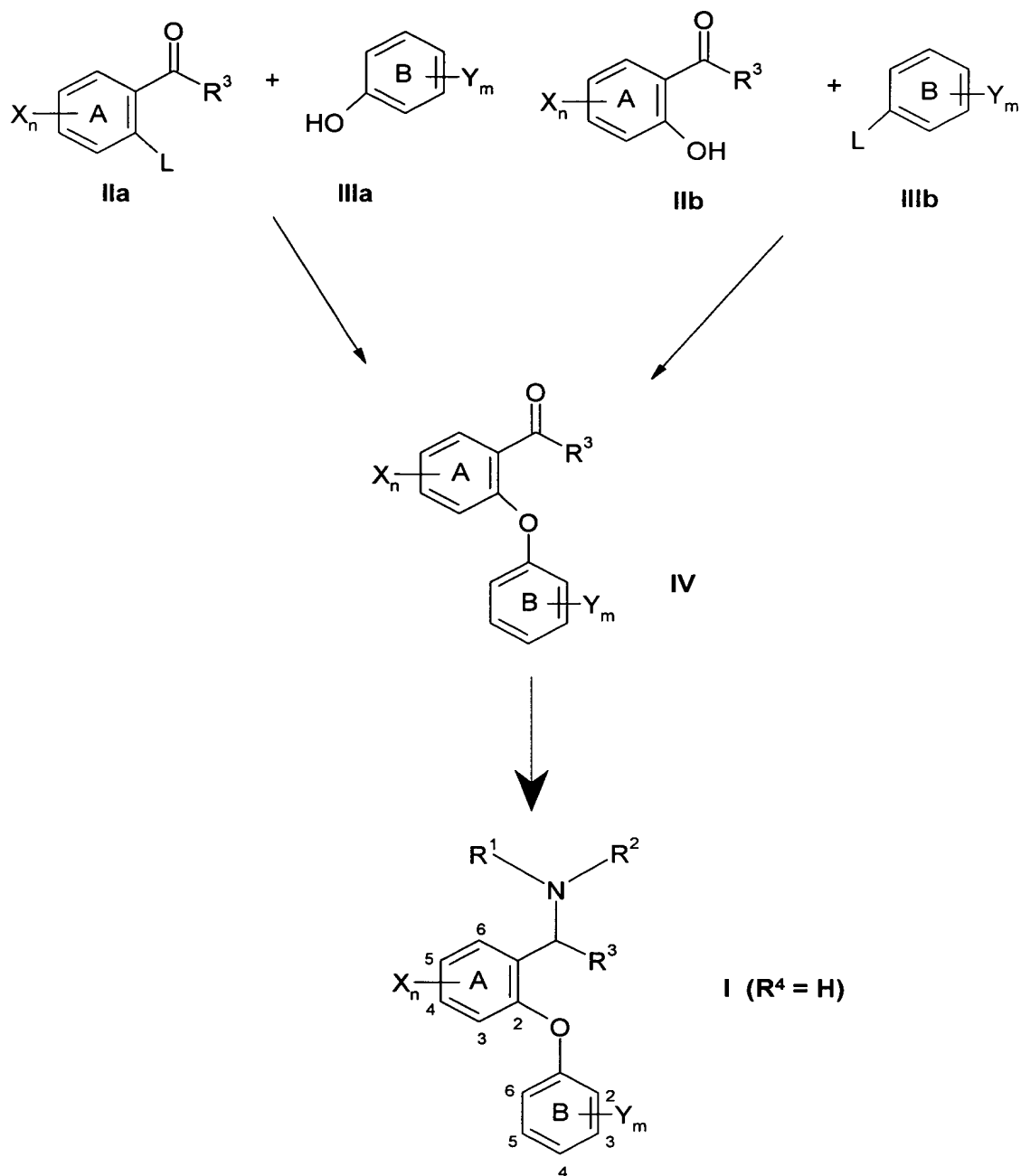
The compounds of formula XVIII may have optical centers and therefore may occur in different enantiomeric configurations. The invention includes all enantiomers, diastereomers, and other stereoisomers of such compounds of formula XVIII, as well as racemic and other mixtures thereof.

Detailed Description of the Invention

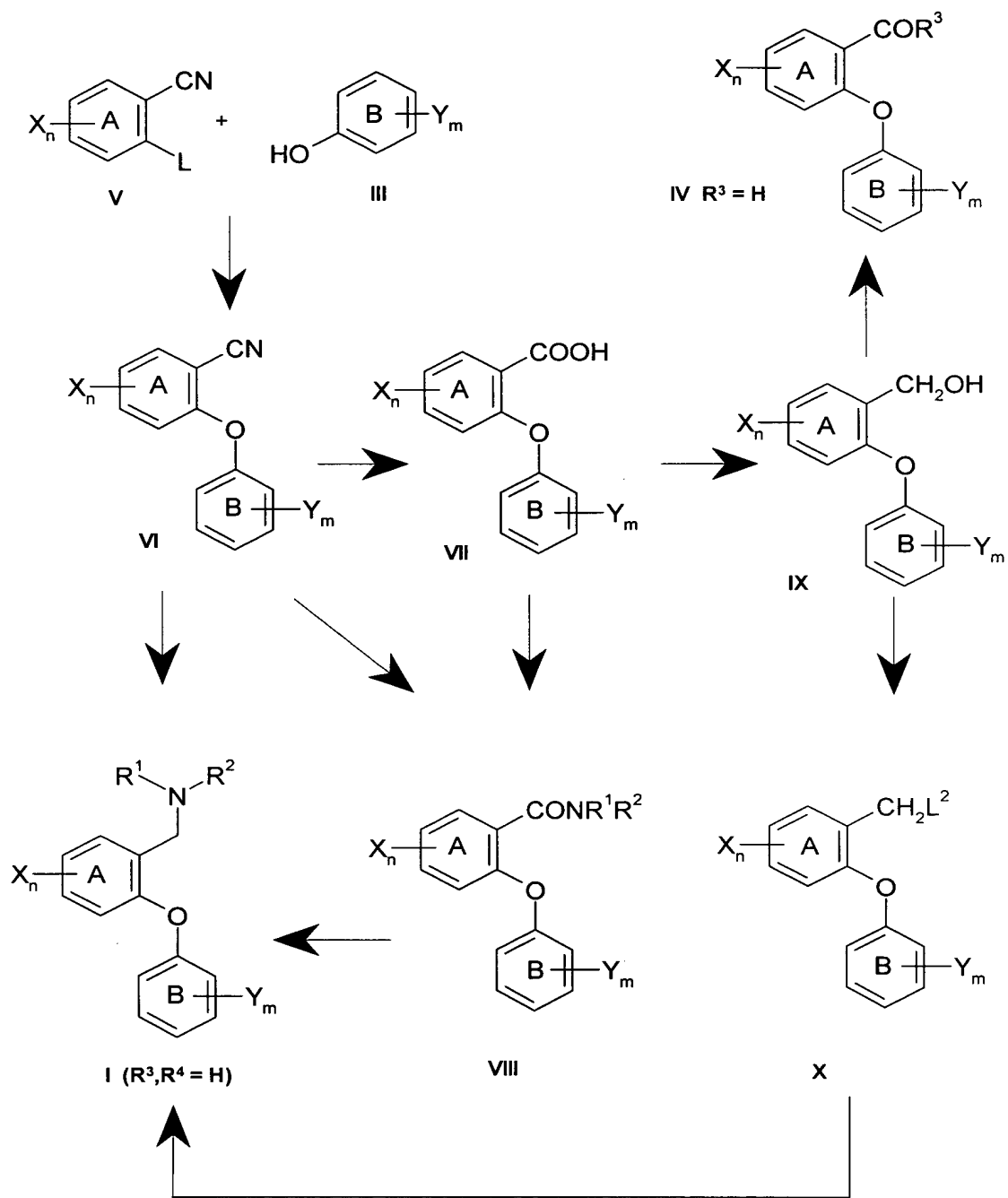
Compounds of the formula I may be prepared according to the following reaction Schemes and discussion. Unless otherwise indicated, A, B, R¹, R², R³, R⁴, R⁵, R⁶, X, Y, m and n, and structural formulas I and XVIII, in the reaction schemes and discussion that follows are as defined above.

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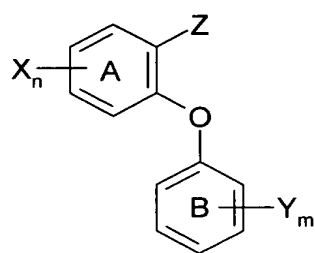
SCHEME 1



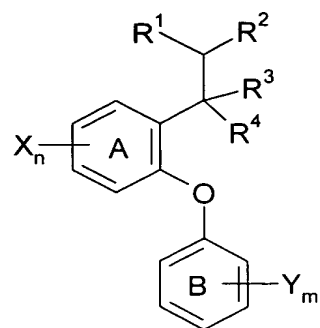
SCHEME 2



SCHEME 3



VI $Z = \text{CN}$
VIII $Z = \text{CONR}^1\text{R}^2$



I (R^3, R^4 not = H)

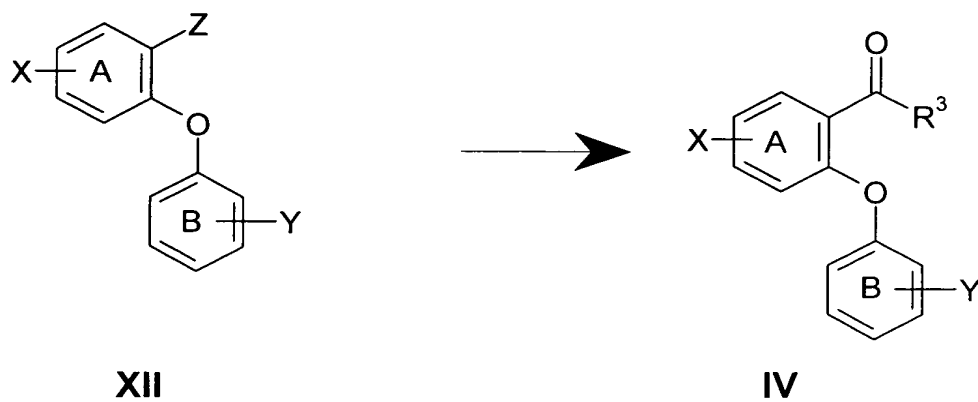
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IV

XI

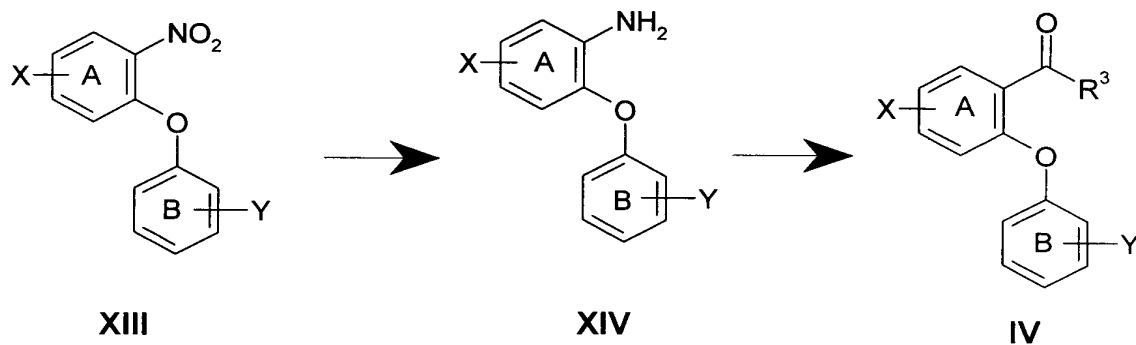
I (R^3, R^4)

SCHEME 5

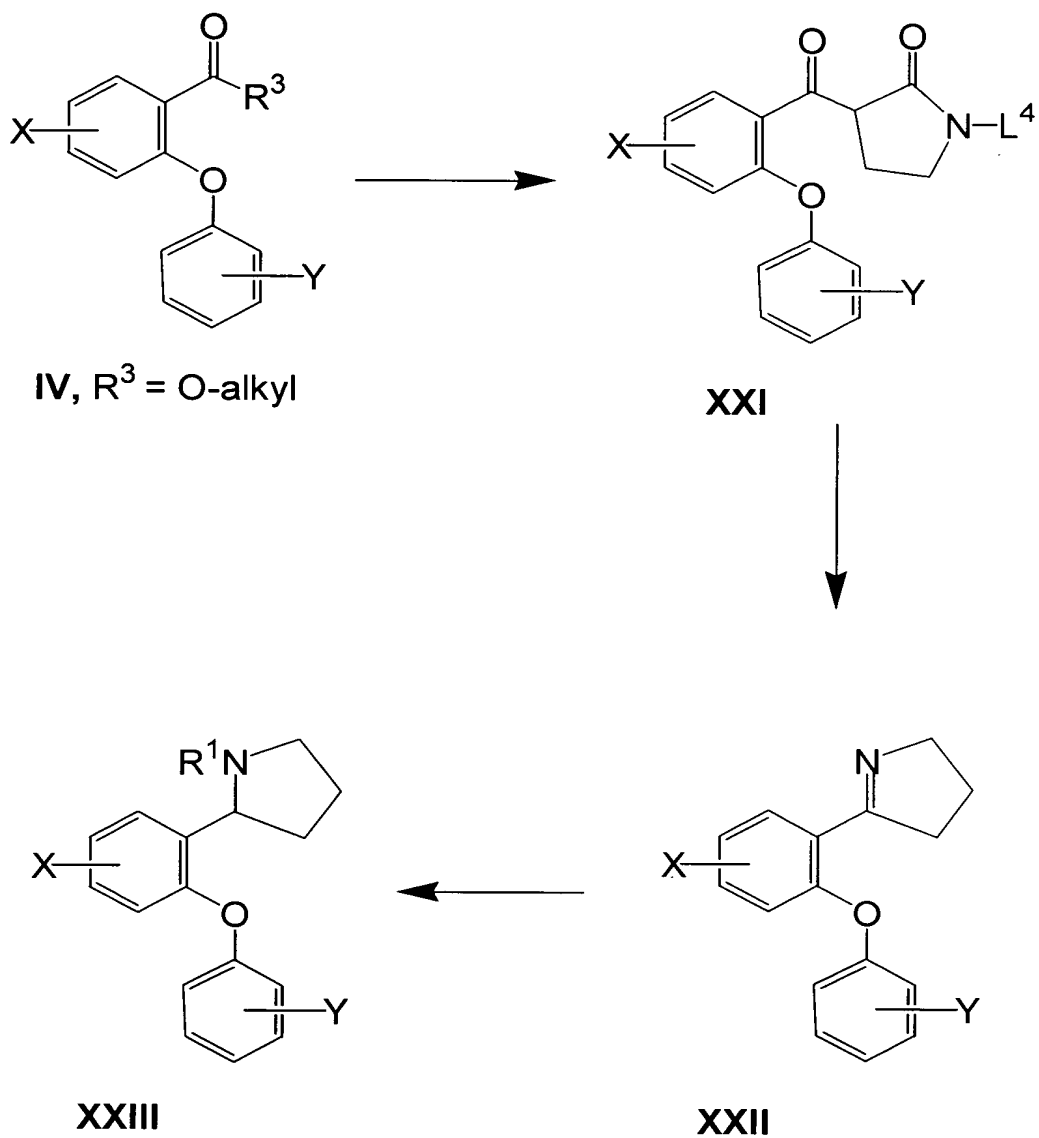


SCHEME 6

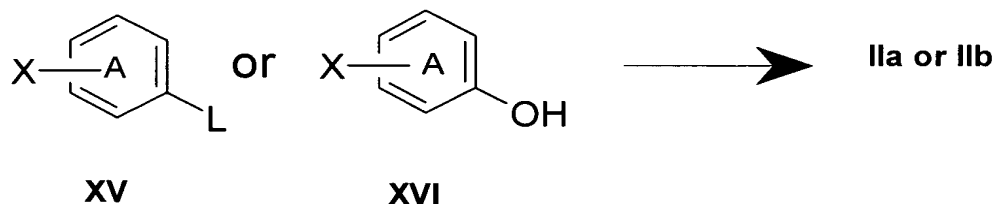
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SCHEME 7



Scheme 1 refers to the preparation of compounds of the formula I from compounds of the formulas II and III. Compounds of the formulas II and III are commercially available or can be made by methods well known to those of ordinary skill in the art. For example, compounds of general formulas IIa and IIb wherein R³ is H may be prepared by introducing an aldehyde functional group (CHO) to a compound of formula XV or XVI, respectively, using methods well known to those of skill in the art.

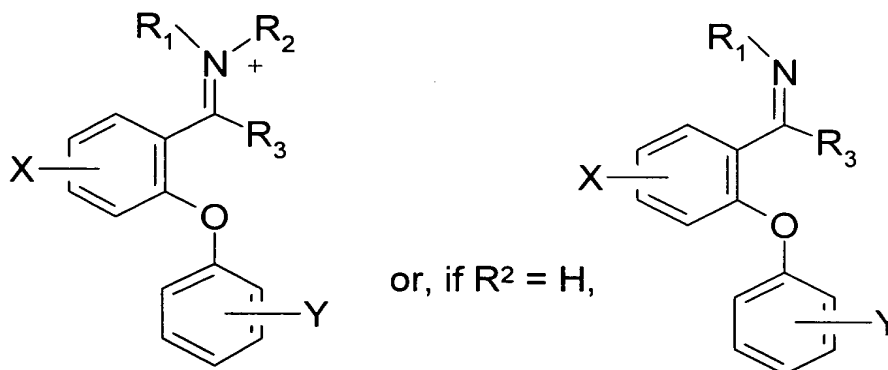


When L = F, the procedure of A. J. Bridges *et al.*, Tetrahedron Letters, **1992**, 33(49), 7499-7502, is particularly useful for the synthesis of substituted ortho-fluorobenzaldehydes. Other such transformations have been described by C. F. H. Allen *et al.*, Organic Synthesis, **1951**, 31, 92; T. DePaulis *et al.*, J. Med. Chem., **1986**, 29, 61; I. M. Godfrey *et al.*, J. Chemical Society, Perkin Transactions 1, **1974**, 1353; K. M. Tramposil *et al.*, J. Med. Chem., **1983**, 26(2), 121; and M. E. Cracknell *et al.*, Chem. Ind., (London), **1985**, (9), 309.

Referring to Scheme 1, a compound (*i.e.*, an aldehyde or ketone) of the formula **IIa**, wherein L is a suitable leaving group such as fluoro, chloro, nitro or triflate, is reacted with a compound (*i.e.*, a phenol) of the formula **IIIa** in the presence of a base to form the corresponding compound of formula **IV**. This reaction is generally carried out at a temperature from about 0°C to about 150°C for about 1 hour to about 3 days, preferably at about 90-95°C for about 18 hours, in a polar solvent such as dimethyl sulfoxide (DMSO), N,N-dimethylformamide (DMF), N,N-dimethylacetamide (DMA) or N-methyl-2-pyrrolidinone (NMP), preferably DMF. Suitable bases include anhydrous sodium carbonate (Na₂CO₃), potassium carbonate (K₂CO₃), sodium hydroxide (NaOH), potassium hydroxide (KOH) and amines such as pyrrolidine, triethylamine and pyridine, with anhydrous K₂CO₃ being preferred. Details for conducting this procedure can be found in G. W. Yeager *et al.*, Synthesis, **1995**, 28-30; J. R. Dimmock *et al.*, Journal of Medicinal Chemistry, **1996**, 39(20), 3984-3997. Alternatively, phenols of the formula **IIb** and compounds of the formula **IIIb** may be converted into aldehydes or ketones of the general formulae **IV** according to the procedures described by K. Tomisawa *et al.*, Chemical and Pharmaceutical Bulletin, **1984**, 32(8), 3066-3074.

Compounds of the formula **IV** can be converted into compounds of the formula **I** by subjecting them to reductive amination conditions. For example, a compound of the formula

IV can be reacted with a compound of the formula HNR^1R^2 to form an intermediate of the formula **XVII**:



XVII

5 which may be isolated or converted directly in the same reaction step into a compound of the formula I. This conversion, whether in situ or starting with the isolated compound of formula XVII, can be performed using one or more methods known to those skilled in the art.

For example, the compound of formula IV and the appropriate compound of formula HNR^1R^2 can be combined in the presence of a dehydrating reagent such as titanium (IV) tetrachloride or titanium (IV) isopropoxide, in a reaction inert solvent such as benzene, toluene, ethanol or a like solvent, until the reaction is judged to be complete, according to the procedure of S. Bhattacharyya (Journal of Organic Chemistry, **1995**, 60(15), 4928-4929). Alternatively, the compound of formula IV and the compound of formula HNR^1R^2 can be combined in an inert solvent such as benzene or toluene, in the presence or absence of a water scavenger such as molecular sieves, and heated to eliminate water generated during the formation of the intermediate of formula XVII. The degree of completion of the conversion of compounds of the formula IV into the above intermediate(s) of formula XVII can be assessed using one or more known analytical techniques, including ^1H -NMR spectroscopy.

In some instances, it may be possible or desirable to isolate the intermediate(s) of formula XVII, or they may be further reacted with a reducing agent selective for the reduction of the intermediate to the desired compounds of formula I. Such reducing agents are widely known to those skilled in the art and include, for example, sodium borohydride (NaBH_4), sodium cyanoborohydride (NaBH_3CN) and sodium triacetoxy-borohydride ($\text{NaBH}(\text{OAc})_3$), as described by A. F. Abdel-Magid et al. in Tetrahedron Letters, **1990**, 31, 5595). This reduction is generally carried out in a polar solvent such as methanol, ethanol, isopropanol or a like

solvent, and at temperatures of about 0°C to about 100°C, preferably at room temperature. In the procedure described by Bhattacharyya, the intermediate of formula **XVII** is formed in an ethanol solvent and, without isolation, is reduced to the product of formula **I** using NaBH₄. As an alternative to the aldehyde or ketone intermediates of formula **IV**, one skilled in the art can also prepare compounds of formula **VI** (i.e., nitriles), as illustrated in Scheme 2, for use as intermediates in the syntheses of the desired compounds of formula **I**, using a diphenyl ether formation procedure analogous to that described in Scheme 1. Procedures for preparation of compounds of formula **VI** can be adapted from those found in the literature. (See, e.g., D. C. Remy *et al.*, *J. Med. Chem.*, **1975**, *18*(2), 142-148; E. A. Schmittling *et al.*, *Journal of Organic Chemistry*, **1993**, *58*(12), 3229-3230).

The conversion of the nitriles of formula **VI** so obtained into the desired products of formula **I** can be achieved by several routes, as depicted in Scheme 2. For example, the nitrile group of **VI** can be hydrolyzed under acidic conditions using methods well known to those of skill in the art, to produce a carboxylic acid derivative of formula **VII**. (See, e.g., A. I. Meyers *et al.*, *Tetrahedron Letters*, **1984**, *25* (28), 2941; and R. W. Higgins *et al.*, *J. Organic Chemistry*, **1951**, *16*, 1275). This carboxylic acid derivative, in turn, can be converted into a carboxamide derivative having the formula **VIII** using one or more standard methods which are disclosed in the chemical literature, e.g., via reaction of an acid halide prepared from compounds of the formula **VII** with an amine of general formula HNR¹R². (See R. E. Kent *et al.*, *Organic Synthesis*, Coll. Vol. III, **1955**, 490; and R. M. Herbst *et al.*, *Organic Synthesis*, Coll. Vol. II, **1943**, 11 for discussions of the Schotten-Bauman reaction). The resulting carboxamides of the formula **VIII**, wherein R¹ and R² are hydrogen, can also be prepared directly from the corresponding nitriles of formula **VI** by specific hydrolysis methods, employing, for example, hydrogen peroxide or strong aqueous alkalis metal salts. (See *Chemistry & Industry*, **1961**, 1987; C. R. Noller, *Organic Synthesis*, Coll. Vol. II, **1943**, 586; and J. H. Hall and M. Gisler, *J. Organic Chemistry*, **1976**, *41*, 3769). Subsequently, the carboxamide derivatives of formula **VIII** may can be reduced using one of a variety of reducing agents available for such conversion, to produce the desired compounds of formula **I**. (See, e.g., A. C. Cope *et al.*, *Organic Synthesis*, Coll. Vol. IV, **1963**, 339, for use of lithium aluminum hydride in a diethyl ether or THF solvent.) Alternatively, the nitriles of formula **VI** can be reduced to form the desired compounds of general formula **I**, wherein R¹ and R² are hydrogen, by using one of a variety of reducing agents disclosed in the chemical literature which are selective for this transformation (including catalytic hydrogenation using hydrogen gas and platinum (II) oxide, as described by J. A. Secrist, III and M. W. Logue in *J. Organic Chemistry*, **1972**, *37*, 335; hydrazine hydrate and Raney nickel in ethanol, as described by W. W. Zajac,

5 Finally, the nitriles of formula **VI** may be converted to the corresponding aldehydes of general formula **IV**, wherein R³ is hydrogen, using reaction conditions specific for this transformation, such as lithium triethoxyaluminum hydride in a solvent such as THF or diethyl ether, as described by H. C. Brown and C. P. Garg in J. American Chemical Society, **1964**, 86, 1085 and by J. Malek and M. Cerny in Synthesis, **1972**, 217.

The alcohols of formula IX so obtained can then be selectively re-oxidized to form the corresponding aldehydes of formula IV ($R^3 = H$), for example, using pyridinium chlorochromate in methylene chloride, according to the procedure of E. J. Corey *et al.*, *Tetrahedron Letters*, **1975**, 31, 2647-2650, or C.-G. Huang, *Journal of Organic Chemistry*, **1991**, 56(16), 4846-4853. The hydroxyl group of the compound of formula IX may be also converted into a more reactive leaving group L^2 (e.g., mesylate, triflate), as described in the scientific literature (see, e.g., M. S. Newman *et al.*, *J. Organic Chemistry*, **1974**, 39, 1036; and E. K. Anderson *et al.*, *Synthesis*, **1974**, 665), to generate intermediates of the formula X, which can then be reacted with the appropriate amines of formula HNR^1R^2 to produce the corresponding desired compounds of formula I.

Alternatively, according to Scheme 4, intermediates of the formula **IV** can be reacted
35 with, for example, a Grignard reagent (i.e., R^4MgJ) under well established conditions to

generate the intermediate alcohols of formula **XI**. Such alcohols of formula **XI** may be converted in one of several manners. For example, the alcohol may be reacted with NaN_3 in aqueous DMF, replacing the OH in formula **XI** with N_3 , according to the procedure of Sharpless *et al.*, Tetrahedron Letters, **1996**, 37(19), 3219-3222. The azide derivative thus
 5 formed can then be reduced to the primary amine of formula **I** ($\text{R}^1=\text{R}^2=\text{H}$) under a variety of conditions, for example, using hydrogen and barium sulfate in ethanol, (A. Guy *et al.*, Synthesis, **1988**, 11, 900-904), lithium aluminum hydride in ether (M. Saito *et al.*, Journal of Medicinal Chemistry, **1980**, 23(12), 1364) or tributyltin hydride in benzene (J. Wasserman *et al.*, Journal of American Chemical Society, **1985**, 107(2), 519). The primary amine ($-\text{NH}_2$)
 10 compound of formula **I** so obtained can be converted to compounds of formula **I** wherein R^1 and R^2 are other than hydrogen, as described previously.

In addition to the methods described above in Schemes 1 and 2 for the preparation of the intermediate aldehydes and ketones of formula **IV**, other methods exist which can provide compounds of the formula **IV**. For example, a compound of formula **XII**, in which the group Z
 15 is a hydrogen atom, can be reacted, under conditions of Friedel-Crafts acylation (*e.g.*, $\text{AlCl}_3/\text{CH}_2\text{Cl}_2/\text{R}^3\text{COCl}$), to produce ketones of the formula **IV** in which R^3 is not hydrogen. This procedure is depicted in Scheme 5 and is well preceded in the scientific literature and familiar to those skilled in the art. The location of the acyl group (COR^3) can be determined by the nature and location of the X and/or Y substituents present, as well as the conditions
 20 employed for the reaction. In instances where it is desirable to prepare compounds of formula **IV** ($\text{R}^3=\text{H}$), from **XII** ($\text{Z}=\text{H}$), introduction of the aldehyde functional group (CHO) may be achieved using conditions described above for the preparation of the intermediates **Ila** and **Ilb** in Scheme 1.

Preparation of compounds of the formula **IV** wherein $\text{R}_3 = \text{H}$ (*i.e.*, aldehydes) can be
 25 achieved using one or more of the known procedures for the formylation of aromatic rings, including reacting dichloromethyl methyl ether and titanium (IV) tetrachloride in methylene chloride according to the procedure described by M. L. Mancini *et al.*, Synthetic Communications, **1989**, 2001-2007, or H. Chikashita *et al.*, J. Organic Chemistry, **1991**, 56, 1692.

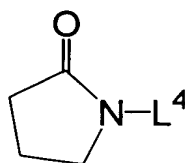
Alternatively, reaction of the compounds of formula **XII** wherein Z is a halogen (*e.g.*, Br, I) with a strong base (*e.g.*, *n*-sec- or *tert*-butyl lithium, lithium diisopropylamide) in an inert solvent such as hexane or THF, followed by reaction with a reagent such as N,N-dimethylformamide (DMF), will produce aldehydes like those of formula **IV**. (See G. Voss *et al.*, Chemische Berichte, **1989**, 122, 1199; M. P. Hoyer *et al.*, J. Organic Chemistry, **1986**,
 30 51(26), 5106; and N. Eisen *et al.*, Angew. Chem. International Edition, **1986**, 25(11), 1026).

Another method for the preparation of compounds of formula **IV** relies on the oxidation of an alkyl group (e.g., CH₃, C₂H₅) at the Z position of compounds of the formula **XII**. The oxidation can progress to the formation of the aldehyde of the formula **IV**, wherein R³ = H (e.g., F. M. Hauser *et al.*, *Synthesis*, **1987**, 723; S. D. Carter *et al.*, *Synthesis*, **1983**, (12), 1000; European Patent 451650, **1991**, Bayer AG), or, under more vigorous conditions, can proceed to the formation of a carboxylic acid compound of formula **VII**, from which it can then be converted into compounds of the formula **I**, as depicted in Scheme 2. The success of this oxidation procedure will, of course, depend upon the nature and location of any additional substituents X and Y on the compounds of formula **XII**.

Yet another alternative for the preparation of the intermediates of formula **IV** is illustrated by Scheme 6. Compounds of the formula **XIII**, containing a nitro group and prepared according to procedures described by R. Beugelmans *et al.* (*Tetrahedron Letters*, **1994**, 35(31), 5649-5652), J. H. Clark *et al.*, (*Tetrahedron Letters*, **1987**, 28(31), 3627) and E. Roberts *et al.*, (*Journal of the Chemical Society*, **1925**, 127, 2004), can be reduced under conditions disclosed in the chemical literature to the corresponding amine compounds of formula **XIV**. These reductions can be accomplished using gaseous hydrogen (H₂) and a catalyst (e.g., Pd/C, Raney nickel) in an alcohol solvent such as ethanol, at pressures of about one to about 5 atmospheres of H₂, or using an *in situ* reduction using an iron/acetic acid or tin/hydrochloric acid system to produce the corresponding compounds of formula **XIV**. These latter intermediates of formula **XIV** can then be converted, via their diazonium salts (prepared, for example, using NaNO₂ and aqueous HCl) into the nitriles of formula **VI** (i.e., see H. T. Clarke and R. R. Read, *Organic Synthesis*, **1941**, 514) which may then be converted via the corresponding carboxylic acids of formula **VII** as shown in Scheme 2 into aldehydes of the general formula **IV** wherein R³ = H.

The intermediates of formula **XIV** can also be converted directly into the corresponding aldehydes of formula **IV** by reacting them with formaldoxime, followed by acid hydrolysis, as described by R.B. Woodward *et al.*, *Tetrahedron*, **1958**, 2, 1 and W. F. Beech in *J. Chemical Society*, **1954**, 1297.

For the preparation of compounds of the general formula **I** wherein R² and R³ taken together with the nitrogen to which R² is attached and the carbon to which R³ is attached form a nitrogen containing ring, an adaptation of the procedure described by L. S. Bleicher *et al.* (*J. Organic Chemistry*, **1998**, 63, 1109) can be employed, as shown in Scheme 7. Thus, an ester of the general formula **IV** (R³ = O-alkyl) is reacted with a cyclic lactam of the general formula **XXX**



XXX

where L^4 is a reaction labile group such as $-CH=CH_2$, in the presence of a strong base such as sodium methoxide, to produce the intermediate of general formula **XXI**. This intermediate
5 can then be converted to the corresponding cyclic imine of formula **XXII** in the presence of a strong acid, such as hydrochloric acid, usually under reflux conditions. Subsequently, the compounds of formula **XXII** can be reduced to form the cyclic amines of formula **XXIII** (wherein $R^1 = H$) using, for example, sodium borohydride in methanol as described previously. Such compounds of formula **XXIII** can further be converted into compounds of the formula
10 **XXIII** (wherein R^1 is as defined for compounds of formula I) as previously discussed.

Pharmaceutically acceptable salts of a compound of formula I can be prepared in a conventional manner by treating a solution or suspension of the corresponding free base or acid with one chemical equivalent of a pharmaceutically acceptable acid or base. Conventional concentration or crystallization techniques can be employed to isolate the salts.
15 Illustrative of suitable acids are acetic, lactic, succinic, maleic, tartaric, citric, gluconic, ascorbic, benzoic, cinnamic, fumaric, sulfuric, phosphoric, hydrochloric, hydrobromic, hydroiodic, sulfamic, sulfonic acids such as methanesulfonic, benzene sulfonic, *p*-toluenesulfonic, and related acids. Illustrative bases are sodium, potassium, and calcium.

A compound of this invention may be administered alone or in combination with
20 pharmaceutically acceptable carriers, in either single or multiple doses. Suitable pharmaceutical carriers include inert solid diluents or fillers, sterile aqueous solutions and various organic solvents. The pharmaceutical compositions formed by combining a compound of formula I or a pharmaceutically acceptable salt thereof can then be readily administered in a variety of dosage forms such as tablets, powders, lozenges, syrups, injectable solutions and the like. These
25 pharmaceutical compositions can, if desired, contain additional ingredients such as flavorings, binders, excipients and the like. Thus, for purposes of oral administration, tablets containing various excipients such as sodium citrate, calcium carbonate and calcium phosphate may be employed along with various disintegrants such as starch, methylcellulose, alginic acid and certain complex silicates, together with binding agents such as polyvinylpyrrolidone, sucrose, gelatin and acacia. Additionally, lubricating agents such as magnesium stearate, sodium lauryl
30 sulfate and talc are often useful for tableting purposes. Solid compositions of a similar type may also be employed as fillers in soft and hard filled gelatin capsules. Preferred materials for this

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glass homogenizing tube with a Teflon™ pestle at 350 rpm using a Potters homogenizer. The homogenate is centrifuged at 1000 x g for 10 min. at 4°C. The resulting supernatant is recentrifuged at 17,000 x g for 20 min. at 4°C. The final pellet is resuspended in an appropriate volume of sucrose buffer that yielded less than 10% uptake.

- 5 Cell Preparation: HEK-293 cells transfected with the human serotonin (5-HT), norepinephrine (NE) or dopamine (DA) transporter are grown in DMEM (Dulbecco's Modified Eagle Medium, Life Technologies Inc., 9800 Medical Center Dr., Gaithersburg, MD, catalog no. 11995-065)) supplemented with 10% dialyzed FBS (Fetal Bovine Serum, from Life Technologies, catalog no. 26300-053), 2 mM L-glutamine and 250 ug/ml G418 for the 5-HT and NE transporter or 2ug/ml puromycin for the DA transporter, for selection pressure. The cells are grown in Gibco triple flasks, harvested with Phosphate Buffered Saline (Life Technologies, catalog no. 14190-136) and diluted to an appropriate amount to yield less than 10% uptake.

- 15 Neurotransmitter Uptake Assay: The uptake assays are conducted in glass tubes containing 50 uL of solvent, inhibitor or 10uM sertraline, desipramine or nomifensine for the 5-HT, NE or DA assay nonspecific uptake, respectively. Each tube contains 400 uL of [3H]5-HT (5 nM final), [3H]NE (10 nM final) or [3H]DA (5 nM final) made up in modified Krebs solution containing 100 uM pargyline and glucose (1mg/ml). The tubes are placed on ice and 50 uL of synaptosomes or cells is added to each tube. The tubes are then incubated at 37° C for 7 min. (5-HT, DA) or 10 min. (NE). The incubation is terminated by filtration (GF/B filters), using a 96-well Brandel Cell Harvester, the filters are washed with modified Krebs buffer and counted using either a Wallac Model 1214 or Wallac Beta Plate Model 1205 scintillation counter.

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EXPERIMENTAL EXAMPLES

PREPARATION 1

2-(3,4-DICHLOROPHENOXY)-5-TRIFLUOROMETHYLBENZALDEHYDE

Under N₂ in a 50 mL round-bottomed flask fitted with a reflux condenser and magnetic stirrer were placed 0.829 g (6.0 mmol) of K₂CO₃ and 0.342 g (2.1 mmol) of 3,4-dichlorophenol (Aldrich Chem. Co., Milwaukee, WI) in 10 mL of anhydrous N,N-dimethylformamide (DMF). After stirring the mixture for 5 min., 0.384 g (2.0 mmol) of 2-fluoro-5-trifluoromethylbenzaldehyde (Aldrich) was added and the mixture was heated to 90-95° C overnight. After allowing the reaction to cool to room temperature, the mixture was diluted with water and EtOAc, the aqueous layer was extracted with additional EtOAc and the organic layers were combined, washed with H₂O and saturated NaCl and dried over MgSO₄. Removal of the solvent in vacuo gave a light amber colored oil, 0.680 g.

¹H-NMR (CDCl₃, 400 MHz): d 10.46 (s, 1H), 8.18 (s, 1H), 7.74 (m, 1H), 7.48 (m, 1H), 7.23 (m, 1H), 6.97 (m, 2H).

Mass spectrum (GCMS, m/z): 334 (m⁺).

In the same manner, the following aldehyde and ketone intermediates of the formula IV were prepared:

| Prep. No. | X _(n) | Y _(m) | R ³ | yield (%) | m.p. (°C) | m/z (m ⁺) | ¹ H-NMR (CDCl ₃ , δ) |
|-----------|------------------|---------------------|----------------|-----------|-----------|-----------------------|--|
| 2 | H | 3,4-Cl ₂ | H | 35 | 57-58 | 269, 267 | (s, 1H), 7.94 (d, 1H), 7.56 (m, 1H), 7.42 (d, 1H), 7.24 (m, 1H), 7.15 (s, 1H), 6.90 (m, 2H). |
| 3 | 5-F | 4-Cl | H | 57 | oil | 252, 250 | (s, 1H), 7.58 (dd, 1H), 7.33 (m, 2H), 7.23 (m, 1H), 6.95 (m, 1H), 6.90 (dd, 1H). |
| 4 | H | 4-Cl | H | 71 | oil | 234, 232 | (s, 1H), 7.92 (dd, 1H), 7.51 (m, 1H), 7.33 (m, 2H), 7.22 (t, 1H), 7.00 (m, 2H), 6.87 (dd, 1H). |

| Prep. No. | X _(n) | Y _(m) | R ³ | yield (%) | m.p. (°C) | m/z (m ⁺) | ¹ H-NMR (CDCl ₃ , δ) |
|-----------|-------------------|--------------------------|-----------------|-----------|-----------|-----------------------|---|
| 5 | 5-F | 3,4-Cl ₂ | H | 67 | oil | 286, 284 | (d, 1H), 7.60 (dd, 1H), 7.42 (d, 1H), 7.28 (m, 1H), 7.10 (d, 1H), 6.95 (dd, 1H), 6.87 (dd, 1H). |
| 6 | 5-F | 3,4-(OCH ₂ O) | H | 38 | oil | 270 | (s, 1H), 7.57 (m, 1H), 7.20 (m, 1H), 7.18 (m, 1H), 7.04 (d, 1H), 6.88 (m, 1H), 6.73 (m, 2H), 2.72 (m, 4H), 1.78 (m, 4H). |
| 7 | 5-CF ₃ | 3,4-Cl ₂ | H | 98 | oil | 336, 334 | (s, 1H), 8.18 (s, 1H), 7.74 (m, 1H), 7.48 (m, 1H), 7.23 (m, 1H), 6.97 (m, 2H). |
| 8 | 5-NO ₂ | 3,4-Cl ₂ | H | 20 | oil | 313, 311 | (s, 1H), 8.78 (d, 1H), 8.35 (dd, 1H), 7.54 (d, 1H), 7.25 (m, 1H), 7.02 (dd, 1H), 6.94 (d, 1H). |
| 9 | 3-CF ₃ | 3,4-Cl ₂ | H | 98 | oil | 336, 334 | (s, 1H), 8.16 (dd, 1H), 7.99 (dd, 1H), 7.54 (t, 1H), 7.34 (d, 1H), 6.92 (d, 1H), 6.6 (dd, 1H). |
| 10 | H | 3,4-Cl ₂ | CH ₃ | 17 | oil | 284, 282 | (dd, 1H), 7.46 (dt, 1H), 7.39 (d, 1H), 7.24 (dt, 1H), 7.21 (d, 1H), 7.08 (d, 1H), 6.92 (dd, 1H), 6.83 (dd, 1H), 2.57 (s, 3H). |

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| Prep. No. | X _(n) | Y _(m) | R ³ | yield (%) | m.p. (°C) | m/z (m ⁺) | ¹ H-NMR (CDCl ₃ , δ) |
|-----------|-------------------|-------------------------------------|-----------------|-----------|-----------|-----------------------|---|
| 11 | H | 4-CH ₃ | H | 69 | oil | 212 | (s, 1H), 7.46 (m, 1H), 7.15 (m, 3H), 6.96 (m, 2H), 6.84 (d, 1H), 2.34 (s, 3H). |
| 12 | 4-CF ₃ | 3,4-Cl ₂ | H | 44 | oil | 338, 336 | (s, 1H), 8.05 (d, 1H), 7.49 (d, 2H), 7.22 (m, 1H), 7.13 (s, 3H), 6.94 (dd, 1H). |
| 13 | H | 3,4-F ₂ | H | 68 | oil | 234 | (s, 1H), 7.93 (dd, 1H), 7.53 (dt, 1H), 7.23 (d, 1H), 7.17 (t, 1H), 6.89 (m, 2H), 6.78 (m, 1H). |
| 14 | H | 4-CH ₃ | CH ₃ | 19 | oil | 227 | (dd, 1H), 7.38 (m, 1H), 7.13 (m, 3H), 6.86 (m, 3H), 2.63 (s, 3H), 2.33 (s, 3H). |
| 15 | 6-CF ₃ | 3,4-Cl ₂ | H | 79 | oil | 336, 334 | (s, 1H), 7.59 (m, 2H), 7.42 (dd, 1H), 7.14 (dd, 1H), 7.12 (m, 1H), 6.87 (m, 1H). |
| 16 | 5-F | 3,4-(CH ₃) ₂ | H | 53 | oil | 244 | (s, 1H), 7.57 (dd, 1H), 7.19 (m, 1H), 7.11 (d, 1H), 6.87 (dd, 1H), 6.81 (m, 1H), 6.75 (dd, 1H), 2.24 (s, 6H). |
| 17 | H | 3,4-(CH ₃) ₂ | H | 60 | oil | 226 | (s, 1H), 7.91 (dd, 1H), |

| Prep. No. | X _(n) | Y _(m) | R ^d | yield (%) | m.p. (°C) | m/z (m ⁺) | ¹ H-NMR (CDCl ₃ , δ) |
|-----------|-------------------|--------------------------|-----------------|-----------|-----------|-----------------------|--|
| | | | | | | | 7.89 (d, 1H), 7.46 (dt, 1H), 7.11 (m, 2H), 6.85 (m, 2H), 6.79 (dd, 1H), 2.24 (s, 6H). |
| 18 | 4-CF ₃ | 3,4-Cl ₂ | H | 37 | oil | 336, 334 | (s, 1H), 8.04 (d, 1H), 7.48 (m, 2H), 7.20 (s, 1H), 7.13 (s, 1H), 6.94 (dd, 1H). |
| 19 | 4-Cl | 3,4-Cl ₂ | H | 55 | oil | 302, 300 | (s, 1H), 7.86 (d, 1H), 7.46 (d, 1H), 7.19 (m, 2H), 6.94 (dd, 1H), 6.86 (d, 1H). |
| 20 | 5-F | 3,4-(OCH ₂ O) | H | 19 | oil | 260 | (s, 1H), 7.56 (dd, 1H), 7.19 (m, 1H), 6.86 (dd, 1H), 6.77 (d, 1H), 6.58 (s, 1H), 6.47 (d, 1H), 5.99 (s, 2H). |
| 21 | 3-F | 3,4-Cl ₂ | H | 63 | foam | 286, 284 | (s 1H), 7.76 (d, 1H), 7.41 (m, 3H), 6.81 (s, 1H), 6.80 (s, 1H). |
| 22 | 5-F | 3,4-Cl ₂ | CH ₃ | 15 | foam | 300, 298 | (dd, 1H), 7.40 (d, 1H), 7.20 (m, 1H), 7.05 (d, 1H), 6.94 (dd, 1H), 6.80 (dd, 1H), 2.55 (s, 3H). |
| 23 | 5-F | 4-Cl | CH ₃ | 27 | foam | 266, 264 | (dd, 1H), 7.30 (d, 2H), 7.15 (dt, 1H), 6.89 (d, 2H), 2.57 (s, 3H). |

| Prep. No. | X _(n) | Y _(m) | R ³ | yield (%) | m.p. (°C) | m/z (m ⁺) | ¹ H-NMR (CDCl ₃ , δ) |
|-----------|--------------------------------------|---------------------|-----------------|-----------|-----------|-----------------------|---|
| 24 | 4-F | 3,4-Cl ₂ | H | 41 | oil | 286, 284 | 10.2 (s, 1H), 7.85 (m, 1H), 7.47 (d, 1H), 7.22 (d, 1H), 6.95 (m, 1H), 6.82 (m, 1H), 6.68 (m, 1H). |
| 25 | 5-OCH ₃ | 3,4-Cl ₂ | H | 9 | oil | 298, 296 | (s, 1H), 7.38 (m, 2H), 7.15 (M, 1h), 7.05 (S, 1h), 6.94 (D, 1h), 6.83 (DD, 1h), 3.84 (S, 3h). |
| 26 | H | 4-Cl | CH ₃ | 29 | oil | 248, 246 | (d, 1H), 7.43 (dt, 1H), 7.31 (s, 1H), 7.29 (s, 1H), 7.21 (t, 1H), 6.92 (d, 1H), 6.88 (d, 1H), 2.60 (s, 3H). |
| 27 | 5-F | 4-Cl | H | 60 | oil | 252, 250 | (s, 1H), 7.59 (dd, 1H), 7.33 (m, 2H), 7.23 (m, 1H), 6.94 (m, 3H). |
| 28 | 4,5-(OCH ₃) ₂ | 3,4-Cl ₂ | H | 40 | oil | 328, 326 | (s, 1H), 7.25 (m, 1H), 7.07 (dd, 1H), 6.95 (dd, 1H), 6.85 (dd, 1H), 6.83 (dd, 1H), 3.92 (s, 3H), 3.85 (s, 3H) |
| 29 | 4,5-(CH ₃) ₂ | 3,4-Cl ₂ | H | 29 | oil | 296, 294 | (s, 1H), 7.67 (s, 1H), 7.38 (d, 1H), 7.07 (s, 1H), 6.85 (m, 1H), 6.71 s, 1H), 2.26 (d, 6H). |
| 30 | 5-Br | 3,4-Cl ₂ | H | 90 | 129-132 | 346, 344 | 10.4 (s, 1H), 8.03 (d, 1H), 7.64 (dd, 1H), 7.45 (dd, 1H), 7.15 (d, 1H), |

| Prep. No. | X _(n) | Y _(m) | R ³ | yield (%) | m.p. (°C) | m/z (m ⁺) | ¹ H-NMR (CDCl ₃ , δ) |
|-----------|------------------|---------------------|----------------|-----------|-----------|-----------------------|---|
| | | | | | | | 6.91 (dd, 1H), 6.89 (dd, 1H) |
| 31 | 4-Br | 3,4-Cl ₂ | H | 47 | solid | 346, 344 | (s, 1H), 7.79 (dd, 1H), 7.47 (m, 1H), 7.37 (m, 1H), 7.19 (m, 1H), 7.03 (m, 1H), 6.94 (m, 1H). |

PREPARATION 32

5-CYANO-2-(3,4-DICHLOROPHENOXY)-BENZALDEHYDE

Under N₂ in a flame-dried 3-neck round bottomed flask fitted with a reflux condenser and magnetic stirrer, a mixture of 5-bromo-2-(3,4-dichlorophenoxy)-benzaldehyde (3.0 g, 8.7 mmol), zinc (II) cyanide (1.5 g, 13 mmol) and tetrakis(triphenylphosphine) palladium (0) (1.5 g, 1.3 mmol) in anhydrous DMF (145 ml) was stirred at room temperature while degassing with N₂ for 5 min. After heating at approximately 80 °C for 90 min., the reaction was judged complete by thin layer chromatography (50% CH₂Cl₂:hexanes) and was allowed to cool to room temperature. The reaction mixture was then diluted with water and ethyl acetate (EA) and stirred another 10 min. The water layer was separated, extracted twice with EA and combined with the original EA layer, and washed with an aqueous solution of Rochelle salt (potassium sodium tartrate tetrahydrate) followed by aqueous NaCl. The organic layer was dried with Na₂SO₄, filtered and concentrated in vacuo to an oil. The oil was flash chromatographed on a 15 X 5 cm column of silica gel, eluting with CH₂Cl₂:hexanes (1:1) to obtain the title product as a white solid, 1.5 g (60%), m.p. 122-126 °C.

Mass spectrum (GC/MS, m/z): 291 (m⁺), 262.

¹H-NMR (CDCl₃, d): 10.47 (s, 1H), 8.22 (d, 1H), 7.75 (dd, 1H), 7.53 (d, 1H), 7.25 (m, 1H), 6.98 (dd, 1H), 6.92 (d, 1H).

In the same manner, **4-cyano-2-(3,4-dichlorophenoxy)-benzaldehyde** was prepared from the corresponding 4-bromo-2-(3,4-dichlorophenoxy)-benzaldehyde, clear oil, 16%. Mass spectrum (GC/MS, m/z): 291 (m⁺). ¹H-nmr (CDCl₃, d): 10.45 (s, 1H), 8.02 (d, 1H), 7.55 (m, 2H), 7.23 (m, 1H), 7.14 (m, 1H), 6.96 (dd, 1H).

PREPARATION 33

4,5-DIMETHOXY-2-FLUOROBENZALDEHYDE

In a flame-dried, round-bottomed flask fitted with a magnetic stirrer and N₂ inlet was placed 4-fluoroveratrole (0.78 g, 5.0 mmol, Aldrich Chemical Co.) in 20 ml of anhydrous

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2-(3,4-DICHLOROPHENOXY)-5-BENZONITRILE

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20

20

20

20

| Prep. No. | X | Y | Yield, % | m/z, m+ | ¹ H-NMR (CDCl ₃ , δ) |
|-----------|---|---|----------|---------|--|
| | | | | | 7.20 (m, 2H), 6.92 (m, 2H). |

PREPARATION 38

2-(3,4-DICHLOROPHENOXY)-5-FLUORO-N-METHYL-BENZAMIDE

A mixture of 0.602 g (2.0 mmol) of 2-(3,4-dichlorophenoxy)-5-fluoro-benzoic acid in 20 mL of anhydrous benzene was treated with 2 drops of N,N-dimethylformamide (DMF), followed by slow addition of 0.785 g (481 μL, 6.6 mmol) of thionyl chloride. After heating to reflux for 3.5 hr, the solution was stirred at room temperature overnight and concentrated in vacuo to a tan oil that was further azeotroped twice with fresh benzene. The residue was re-dissolved in 4 mL of benzene and added to 15 mL of 1.0 M CH₃NH₂ in CH₃OH (Aldrich Chem. Co.) at 25 °C. After 12 hr, the mixture was concentrated in vacuo and partitioned between water and EtOAc. The aqueous layer was then extracted with additional EtOAc and the combined organic extracts were washed with water followed by saturated aqueous NaCl, dried with MgSO₄, filtered and concentrated to a brown oil which slowly solidified on standing, 0.654 g.

¹H-nmr (CDCl₃, 400 MHz, δ): 7.89 (dd, 1H), 7.41 (d, 1h), 7.30 (bs, 1H, NH), 7.09 (m, 2H), 6.83 (m, 2H), 2.95 (d, 3H).

PREPARATION 39

2-(3,4-DICHLOROPHENOXY)-5-FLUORO-BENZONITRILE

A flame-dried flask containing N₂ inlet and magnetic stirrer was charged with 2.9 g (18 mmol) of 3,4-dichlorophenol, 6.2 g (45 mmol) of potassium carbonate and 50 mL of anhydrous DMF. With stirring, 2.08 g (15 mmol) of 2,5-difluorobenzonitrile (Aldrich Chem. Co.) was added and the mixture was heated overnight at 105 °C, cooled to room temperature and partitioned between water and EtOAc. The aqueous layer was reextracted with EtOAc and the combined organic layers were successively washed with 2N NaOH, water and saturated aqueous NaCl. After drying with MgSO₄, the solvent was removed to give the title product as a light brown oil which slowly solidified, 3.5 g.

Mass spectrum (APCI, m/z): 281 (m⁺).

¹H-nmr (CDCl₃, 400 MHz, δ): 7.43 (d, 1H), 7.37 (dd, 1H), 7.26 (m, 1H), 7.12 (d, 1H), 6.92 (m, 2H).

PREPARATION 40

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Further elution of the chromatography provided the more polar component, **2-(p-tolyloxy)benzylamine**, as a yellow oil, 0.082 g.

¹H-NMR (CDCl₃, 400 MHz): d 7.34 (d, 1H), 7.12 (m, 4H), 6.86 (m, 3H), 3.87 (s, 2H), 2.31 (s, 3H), 1.80 (bs, 2H, NH₂).

- 5 By the same method, the following benzylamino compounds of formula I (where R¹⁻⁴ = H) were prepared from the corresponding benzonitriles VI:

| Ex. No. | X | Y | yield(%) | m.p., °C (HCl salt) | m/z, m ⁺ | ¹ H-NMR (DMSO-d ₆ , δ) |
|---------|--------------------|---------------------|-----------|---------------------|---------------------|---|
| 2 | H | 3,4-Cl ₂ | 87 | 198-201 | 270, 268 | (bs, 1H, HCl), 7.65 (d, 1H), 7.59 (d, 1H), 7.36 (m, 2H), 7.22 (t, 1H), 7.06 (dd, 1H), 6.92 (d, 1H), 4.00 (s, 2H). |
| 3 | 5-F | 3,4-Cl ₂ | 67 | 190-194 | 288, 286 | (d, 1H), 7.52 (dd, 1H), 7.31 (s, 1H), 7.23 (dt, 1H), 7.03 (m, 2H), 3.99 (bs, 2H). |
| 4 | 5-CH ₃ | 3,4-Cl ₂ | 54 | 217-219 | 284, 282 | (d, 1H), 7.38 (s, 1H), 7.22 (d, 1H), 7.20 (m, 1H), 6.99 (m, 1H), 6.87 (d, 1H), 3.94 (bs, 2H), 2.28 (s, 3H). |
| 5 | 4-OCH ₃ | 3,4-Cl ₂ | 37 | 167-168 | 300, 298 | (d, 1H), 7.52 (d, 1H), 7.35 (d, 1H), 7.06 (dd, 1H), 6.82 (dd, 1H), 6.45 (d, 1H), 3.90 (m, 2H), 3.68 (s, 3H). |

EXAMPLE 6

**[2-(3,4-DICHLOROPHENOXY)-5-TRIFLUOROMETHYLBENZYL]-DIMETHYLAMINE
MALEATE**

In a round-bottomed flask fitted with a magnetic stirrer and N₂ inlet was placed 0.331 g (4.06 mmol) of dimethylamine hydrochloride (Aldrich) and 0.410 g (4.06 mmol) triethylamine in 25 mL ethanol while stirring until the solution was clear. At room temperature, 1.15 g titanium (IV) isopropoxide (1.2 mL, 4.06 mmol) was added via syringe, followed by 0.680 g (2.03 mmol) of 2-(3,4-dichlorophenoxy)-5-trifluoromethyl-benzaldehyde to give a yellow-brown solution which was stirred overnight. To the resulting cloudy solution was added 0.115 g (3.05 mmol) of sodium borohydride, and stirring was continued for 24 hr. The reaction was then quenched with 6N HCl (~pH = 10), stirred another 2 hr and diluted with EtOAc. The aqueous layer was made basic with saturated aqueous Na₂CO₃, layered with additional EtOAc and the two-layered mixture was filtered through a pad of diatomaceous earth (d.e.), washing the d.e. pad well with EtOAc and H₂O. The EtOAc layer was combined with additional extracts of the water layer and the combined organics were washed with saturated aqueous NaCl, dried with MgSO₄ and concentrated in vacuo to a yellow oil, 0.626 g.

Mass spectrum: (APCI, m/z) 366, 364 (m⁺).

¹H-NMR (CDCl₃, 400 MHz): d 7.78 (s, 1H), 7.47 (dd, 1H), 7.39 (s, 1H), 7.04 (s, 1H), 6.90 (dd, 1H), 6.80 (dd, 1H), 3.48 (s, 2H), 2.26 (s, 6H).

The preceding oil dissolved in anhydrous diethyl ether was treated with 0.199 g of maleic acid in 2 mL of acetone. The mixture was filtered through a thin pad of d.e. to clarify the solution and then was stirred at room temperature for 8 hr. The resulting solids were filtered and washed with Et₂O and dried under vacuum, m.p 127-128 °C.

Elemental analysis for C₁₆H₁₄Cl₂F₃NO•C₄H₄O₄ calculated: C, 50.01, H, 3.77, N, 2.92.

Found: C, 49.93, H, 3.85, N, 3.04.

In the same manner the following compounds of formula I were prepared:

| Ex. No. | X | Y | R ³ | NR ¹ R ² | mp, °C | m/z, m ⁺ | Elemental Analysis formula CHN calculated:CHN found |
|---------|---|---------------------|----------------|----------------------------------|---------|---------------------|---|
| 7 | H | 3,4-Cl ₂ | H | N(CH ₃) ₂ | 173-174 | 298, 296 | C ₁₅ H ₁₅ Cl ₂ NO•HCl•H ₂ O C 54.16, H 4.85, N 4.21: C 54.02, H 4.77, N 4.23 |

| Ex. No. | X | Y | R ³ | NR ¹ R ² | mp, °C | m/z, m ⁺ | Elemental Analysis formula CHN calculated: CHN found |
|---------|-----|---------------------|----------------|----------------------------------|-------------|---------------------|---|
| 8 | H | 3,4-Cl ₂ | H | NHCH ₃ | 126 | 284, 282 | C ₁₄ H ₁₃ Cl ₂ NO• ^{3/4} C ₄ H ₄ O ₄ C 54.42, H 4.48, N 3.73: C 54.46, H 4.48, N 3.52 |
| 9 | H | 3,4-Cl ₂ | H | NHC ₂ H ₅ | 133- 135 | 298, 296 | C ₁₅ H ₁₅ Cl ₂ NO• C ₄ H ₄ O ₄ C 55.35, H 4.65, N 3.40: C 55.16, H 4.68, N 3.40 |
| 10 | 5-F | 4-Cl | H | NHCH ₃ | 208- 209 | 268, 266 | C ₁₄ H ₁₃ ClFNO•HCl C 55.64, H 4.66, N 4.63: C 55.64, H 4.52, N 4.55 |
| 11 | H | 4-Cl | H | N(CH ₃) ₂ | 122 | 264, 262 | C ₁₅ H ₁₆ ClNO• C ₄ H ₄ O ₄ C 60.40, H 5.33, N 3.70: C 60.46, H 5.29, N 3.71 |
| 12 | H | 4-Cl | H | NHCH ₃ | 128- 129 | 250, 248 | C ₁₄ H ₁₄ ClNO•C ₄ H ₄ O ₄ •1/3H ₂ O C 58.46, H 5.09, N 3.79: C 58.26, H 4.87, N 3.89 |
| 13 | 5-F | 3,4-Cl ₂ | H | N(CH ₃) ₂ | 174- 175 | 316, 314 | C ₁₅ H ₁₄ Cl ₂ FNO•HCl C 51.38, H 3.89, N 4.16: C 51.35, H 4.22, N 3.91 |
| 14 | 5-F | 3,4-Cl ₂ | H | NHCH ₃ | 202- 203 | 302, 300 | C ₁₄ H ₁₂ Cl ₂ FNO•HCl C 49.95, H 3.89, N 4.16: C 50.05, H |

| Ex. No. | X | Y | R ³ | NR ¹ R ² | mp, °C | m/z, m ⁺ | Elemental Analysis formula CHN calculated: CHN found |
|---------|-------------------|---------------------|-----------------|----------------------------------|---------|---------------------|--|
| | | | | | | | 4.04, N 4.21 |
| 15 | 5-F | 3,4-Cl ₂ | H | NHCH ₃ | 129-130 | 302, 300 | C ₁₄ H ₁₂ Cl ₂ FNO • ½ C ₄ H ₄ O ₄ C 50.48, H 4.37, N 3.68: C 50.24, H 4.37, N 3.68 |
| 16 | 5-NO ₂ | 3,4-Cl ₂ | H | N(CH ₃) ₂ | | 343, 341 | Not determined |
| 17 | 3-CF ₃ | 3,4-Cl ₂ | H | N(CH ₃) ₂ | 229-230 | 366, 364 | C ₁₆ H ₁₄ Cl ₂ F ₃ NO • HCl • H ₂ O C 45.90, H 4.09, N 3.35: C 45.70, H 3.82, N 3.30 |
| 18 | H | 3,4-Cl ₂ | CH ₃ | N(CH ₃) ₂ | 130-133 | 312, 310 | C ₁₆ H ₁₇ Cl ₂ NO • HCl Not determined |
| 19 | H | 4-CH ₃ | H | N(CH ₃) ₂ | 111-113 | 242 | C ₁₆ H ₁₉ NO • C ₄ H ₄ O ₄ • ¼ H ₂ O C 66.38, H 6.55, N 3.87: C 66.47, H 6.40, N 3.95 |
| 20 | 4-CF ₃ | 3,4-Cl ₂ | H | N(CH ₃) ₂ | 131-133 | 366, 364 | C ₁₆ H ₁₄ Cl ₂ F ₃ NO • C ₄ H ₄ O ₄ • 2/3 H ₂ O C 48.80, H 3.96, N 2.85: C 48.85, H 3.77, N 2.97 |
| 21 | H | 3,4-F ₂ | H | N(CH ₃) ₂ | 145-147 | 264 | C ₁₅ H ₁₅ F ₂ NO • C ₄ H ₄ O ₄ C 60.15, H 5.05, N 3.69: C 60.09, H |

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| Ex. No. | X | Y | R ³ | NR ¹ R ² | mp, °C | m/z, m ⁺ | Elemental Analysis formula CHN calculated: CHN found |
|---------|-------------------|-------------------------------------|-----------------|----------------------------------|---------|---------------------|---|
| | | | | | | | 4.91, N 3.74 |
| 22 | H | 3,4-F ₂ | H | NHCH ₃ | 101-102 | 250 | C ₁₄ H ₁₃ F ₂ NO• C ₄ H ₄ O ₄ C 59.17, H 4.69, N 3.83: C 59.04, H 4.82, N 3.93 |
| 23 | H | 4-CH ₃ | CH ₃ | NHCH ₃ | 114-116 | 242 | C ₁₆ H ₁₉ NO•C ₄ H ₄ O ₄ •1/8H ₂ O C 66.79, H 6.52, N 3.89: C 66.82, H 6.58, N 3.95 |
| 24 | 6-CF ₃ | 3,4-Cl ₂ | H | N(CH ₃) ₂ | 166-168 | 366, 364 | C ₁₆ H ₁₄ Cl ₂ F ₃ NO•HCl C 47.96, H 3.77, N 3.49: C 47.95, H 3.81, N 3.48 |
| 25 | 6-CF ₃ | 3,4-Cl ₂ | H | NHCH ₃ | 123-125 | 352, 350 | C ₁₅ H ₁₂ Cl ₂ F ₃ NO• C ₄ H ₄ O ₄ C 48.94, H 3.45, N 3.00: C 48.90, H 3.58, N 3.23 |
| 26 | 5-F | 3,4-(CH ₃) ₂ | H | NHCH ₃ | 123-124 | 260 | C ₁₆ H ₁₈ FNO•C ₄ H ₄ O ₄ C 63.99, H 5.91, N 3.73: C 63.89, H 5.91, N 3.760 |
| 27 | H | 3,4-(CH ₃) ₂ | H | NHCH ₃ | 118-119 | 242 | C ₁₆ H ₁₉ NO•C ₄ H ₄ O ₄ C 67.20, H 6.48, N 3.91: C 66.91, H 6.48, N 3.96 |

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| Ex. No. | X | Y | R ³ | NR ¹ R ² | mp, °C | m/z, m ⁺ | Elemental Analysis formula CHN calculated: CHN found |
|---------|-------------------|--------------------------|-----------------|--------------------------------|---------|---------------------|---|
| 28 | 4-CF ₃ | 3,4-Cl ₂ | H | NHCH ₃ | 164-166 | 352, 350 | C ₁₅ H ₁₂ Cl ₂ F ₃ NO• C ₄ H ₄ O ₄ C 48.94, H 3.46, N 3.00: C 49.02, H 3.22, N 3.06 |
| 29 | 4-Cl | 3,4-Cl ₂ | H | NHCH ₃ | 132-133 | 318, 316 | C ₁₄ H ₁₂ Cl ₃ NO• C ₄ H ₄ O ₄ C 49.96, H 3.72, N 3.24: C 49.89, H 3.49, N 3.25 |
| 30 | 5-F | 3,4-(OCH ₂ O) | H | NHCH ₃ | 96-97 | 276 | C ₁₅ H ₁₄ NO ₃ • C ₄ H ₄ O ₄ C 58.31, H 4.63, N 3.57: C 58.34, H 4.38, N 3.62 |
| 31 | 3-F | 3,4-Cl ₂ | H | NHCH ₃ | 138-140 | 302, 300 | C ₁₄ H ₁₂ Cl ₂ FNO• C ₄ H ₄ O ₄ C 51.94, H 3.87, N 3.36: C 51.96, H 3.87, N 3.44 |
| 32 | 5-F | 3,4-Cl ₂ | CH ₃ | NHCH ₃ | 127 | 316, 314 | C ₁₅ H ₁₄ Cl ₂ FNO• C ₄ H ₄ O ₄ C 53.04, H 4.22, N 3.25: C 53.23, H 4.21, N 3.23 |
| 33 | 5-F | 4-Cl | CH ₃ | NHCH ₃ | 173-175 | 282, 280 | C ₁₅ H ₁₅ ClFNO•HCl C 56.98, H 5.10, N 4.43: C 56.90, H 5.34, N 4.42 |
| 34 | 4-F | 3,4-Cl ₂ | H | NHCH ₃ | 151-152 | | C ₁₄ H ₁₂ Cl ₂ FNO• C ₄ H ₄ O ₄ C 50.31, H 4.10, N 3.26: C |

| Ex. No. | X | Y | R ³ | NR ¹ R ² | mp, °C | m/z, m ⁺ | Elemental Analysis formula CHN calculated: CHN found |
|---------|--------------------------------------|---------------------|-----------------|----------------------------------|---------|---------------------|--|
| | | | | | | | 50.28, H 3.93, N 3.24 |
| 35 | 5-OCH ₃ | 3,4-Cl ₂ | H | NHCH ₃ | 140-141 | 314, 312 | C ₁₅ H ₁₅ Cl ₂ NO ₂ •HCl•1/4 H ₂ O C 51.01, H 4.71, N 3.97: C 51.08, H 4.54, N 3.91 |
| 36 | H | 4-Cl | CH ₃ | NHCH ₃ | 147-149 | 264, 262 | C ₁₅ H ₁₆ ClNO•HCl C 60.41, H 5.74, N 4.69: C 60.02, H 5.74, N 4.65 |
| 37 | 5-F | 4-Cl | H | N(CH ₃) ₂ | 103 | 282, 280 | C ₁₅ H ₁₅ ClFNO•C ₄ H ₄ O ₄ C 57.65, H 4.83, N 3.53: C 57.74, H 4.82, N 3.47 |
| 38 | 4,5-(OCH ₃) ₂ | 3,4-Cl ₂ | H | NHCH ₃ | 143-145 | 344, 342 | C ₁₆ H ₁₇ Cl ₂ NO ₃ •HCl C 47.37, H 5.20, N 3.45: C 47.36, H 4.93, N 3.49 |
| 39 | 4,5-(CH ₃) ₂ | 3,4-Cl ₂ | H | NHCH ₃ | | 312, 310 | C ₁₆ H ₁₇ Cl ₂ NO•HCl C 55.43, H 5.23, N 4.04: C 55.84, H 4.80, N 3.80 |
| 40 | 5-Br | 3,4-Cl ₂ | H | NHCH ₃ | 206-210 | 364, 362 | C ₁₄ H ₁₂ BrCl ₂ NO•HCl C 42.30, H 3.30, N 3.52: C 42.13, H 3.13, N 3.36 |
| 41 | 4-Br | 3,4-Cl ₂ | H | NHCH ₃ | 185-189 | 364, 362 | C ₁₄ H ₁₂ BrCl ₂ NO•HCl C 42.30, H 3.30, N 3.52: C 42.03, H |

| Ex. No. | X | Y | R ³ | NR ¹ R ² | mp, °C | m/z, m ⁺ | Elemental Analysis formula CHN calculated: CHN found |
|---------|---|---------------------|----------------|----------------------------------|---------|---------------------|--|
| | | | | | | | 3.07, N 3.33 |
| 42 | H | 4-SCH ₃ | H | N(CH ₃) ₂ | 171-173 | 274 | C ₁₆ H ₁₉ NOS•HCl Not determined |
| 43 | H | 3,4-Cl ₂ | H | NHCH ₃ | 162-164 | 298, 296 | C ₁₅ H ₁₅ Cl ₂ NO•HCl C 54.32, H 4.55, N 4.22: C 54.05, H 4.76, N 4.17 |
| 44 | 4-CN | 3,4-Cl ₂ | H | NHCH ₃ | 200-205 | 309, 307 | C ₁₅ H ₁₂ Cl ₂ N ₂ O•HCl• ¼ H ₂ O C 51.75, H 3.91, N 8.05: C 51.82, H 3.80, N 7.83 |
| 45 | 5-SO ₂ ⁻ NHCH ₃ | 3,4-Cl ₂ | H | NHCH ₃ | 202-204 | 376, 374 | C ₁₅ H ₁₆ Cl ₂ N ₂ O ₃ S•HCl C 43.75, H 4.16, N 6.80: C 43.66, H 3.88, N 6.37 |
| 46 | 4-OCH ₃ | 3,4-Cl ₂ | H | N(CH ₃) ₂ | 113-115 | 328, 326 | C ₁₆ H ₁₇ Cl ₂ NO ₂ • C ₄ H ₄ O ₄ C 54.31, H 4.78, N 3.17: C 54.34, H 4.59, N 2.97 |
| 47 | 5-CH ₃ | 3,4-Cl ₂ | H | N(CH ₃) ₂ | 196-198 | 312, 310 | C ₁₆ H ₁₇ Cl ₂ N•HCl C 55.43, H 5.23, N 4.04: C 55.28, H 5.12, N 4.00 |

EXAMPLE 52**(2-FLUORO-6-(P-TOLYLOXY)BENZYL)-DIMETHYLAMINE MALEATE (I, R^{1,2} = CH₃)**

Under N₂, a solution of 0.120 g (0.52 mmol) of 2-fluoro-6-(p-tolyloxy)benzylamine
 5 (from example 3) in 2 mL of CH₃OH was treated with 215 mL (2.86 mmol) of 37% aqueous formaldehyde (Aldrich), producing a solid precipitate. Sodium triacetoxyborohydride (0.319 g,

1.43 mmol) was then added, producing some foaming before becoming a clear solution. After stirring at room temperature overnight, H₂O and EtOAc were added, the aqueous layer was further extracted with EtOAc, and the organic layers were combined and washed with H₂O and saturated NaCl. After drying over MgSO₄, the solvent was removed in vacuo to give a light yellow film, 0.110 g.

Mass spectrum (APCI, m/z): 260 (m⁺).

¹H-NMR (CDCl₃, 400 MHz): δ 7.28 (m, 1H), 7.14 (m, 2H), 6.91 (m, 2H), 6.83 (t, 1H), 6.56 (d, 1H), 4.38 (s, 2H), 2.79 (s, 6H), 2.32 (s, 3H).

The crude amine (84 mg) was dissolved in Et₂O and treated with 38 mg (1 equiv.) of maleic acid in 2 mL of acetone. After stirring at room temperature for 18 hr, the white solid salt was filtered and washed with small amounts of Et₂O and dried under high vacuum to give 0.048 g of the title product, m.p. 147-148 °C.

Elemental analysis for C₁₆H₁₈FNO•C₄H₄O₄ calculated: C, 63.99, H, 5.90, N, 3.73. Found: C, 63.97, H, 5.91, N, 3.67.

EXAMPLE 53

4-(3,4-DICHLOROPHENOXY)-3-(DIMETHYLAMINOMETHYL)-PHENYLAMINE

HYDROCHLORIDE

Under N₂, in a 250 mL Parr apparatus bottle, 0.265 g of 10% Pd on carbon was treated with 30 mL of EtOAc and 0.53 g (1.55 mmol) of [2-(3,4-dichlorophenoxy)-5-nitrobenzyl]-dimethylamine (prepared as in Preparation 1) followed by 2 drops of acetic acid. The mixture was hydrogenated at 40-45 psi of H₂ for 90 min, at which time no starting material was visible by tlc (90:10 CHCl₃:CH₃OH). Saturated NaHCO₃ was added to adjust the pH>7 and the solution was filtered through a pad of d.e., washing the pad with H₂O and EtOAc. The organic layers were combined, washed with H₂O and saturated NaCl, dried over MgSO₄ and concentrated in vacuo to a tan oil, 0.325 g.

Mass spectrum (APCI, m/z): 313, 311.

¹H-NMR (CDCl₃, 400 MHz): δ 7.39 (m, 2H), 7.21 (m, 3H), 6.70 (dd, 1H), 4.52 (s, 2H), 2.89 (s, 6H).

The oil was dissolved in Et₂O and treated with 2.1 mL of 1.0 M HCl in Et₂O (Aldrich), stirred at room temperature for 2 hr. and filtered. After drying in vacuo, the title product weighed 0.286 g, m.p. 228 °C.

EXAMPLE 54

N-[4-(3,4-DICHLOROPHENOXY)-3-DIMETHYLAMINOMETHYLPHENYL]-ACETAMIDE

HYDROCHLORIDE

To a suspension of 0.130 g (0.375 mmol) of 4-(3,4-dichlorophenoxy)-3-(dimethylaminomethyl)-phenylamine hydrochloride (from example 53) in 3 mL of toluene at

room temperature, under N₂, was added 110 mL of triethylamine. After 2 hr, 28 mL (0.394 mmol) of acetyl chloride was added and the mixture was stirred another 1 hr before being diluted with EtOAc and saturated aqueous NaHCO₃. The aqueous layer was further extracted with EtOAc and the organic layers were combined, washed with H₂O and saturated NaCl, dried with MgSO₄ and concentrated to a tan oil, 0.116 g.

Mass spectrum (APCI, m/z): 355, 353.

The crude product was dissolved in Et₂O and treated with 330 mL of 1.0 M HCl in Et₂O. After stirring at room temperature, the solids were filtered and washed with small amounts of Et₂O, then dried in vacuo to give the title product, 88 mg, m.p. 199-202 °C.

Elemental Analysis calculated for C₁₇H₁₈Cl₂N₂O₂·HCl·³/₄ H₂O: C, 50.64, H, 5.12, N, 6.95. Found: C, 50.51, H, 5.19, N, 6.66.

EXAMPLE 55

2-[2-(3,4-DICHLOROPHENOXY)-5-FLUOROPHENYL]-PYRROLIDINE HYDROCHLORIDE

A. 2-(3,4-DICHLOROPHENOXY)-5-FLUORO-BENZOIC ACID

Under N₂ in a round-bottomed flask fitted with a reflux condenser and magnetic stirrer were placed 6.37 g (19.55 mmol) of cesium carbonate and 3.2 g (19.55 mmol) of 3,4-dichlorophenol (both from Aldrich Chem. Co., Milwaukee, WI) in 60 mL of anhydrous toluene. After stirring the mixture for 5 min., 89 mg (0.24 mmol) of copper (II) trifluoromethanesulfonate (copper triflate) and 0.26g (9.78 mmol) of 5-fluoro-2-iodobenzoic acid (prepared according to the method in Collection of Czechoslovakian Chemical Communications, 1975, vol. 40, p728) were added and the mixture was heated to reflux overnight. Progress of the reaction was monitored using thin layer chromatography (tlc), eluting with CHCl₃:CH₃OH:AcOH (9:1:0.5). After allowing the reaction to cool to room temperature, the mixture was diluted with water and EtOAc; the aqueous layer was acidified with 6 N HCl and re-extracted with additional EtOAc. The organic layers were combined, washed with H₂O and saturated NaCl and dried over MgSO₄. Removal of the solvent in vacuo gave a dark brown colored oil, 2.6 g, as a mixture of title product and unreacted 5-fluoro-2-iodobenzoic acid.

Alternatively, a solution of 4.28 g (15 mmol) of 2-(3,4-dichloro-phenoxy)-5-fluorobenzaldehyde (Preparation no. 5) in 25 mL of acetone was cooled to 5-10 °C and treated via syringe with 5.8 mL (15.6 mmol) of 2.67 M Jones reagent *. After 1 hr at this temperature, the reaction was quenched with 8 mL of isopropanol, allowed to warm to 25 °C, and filtered through a pad of d.e. The filtrate was concentrated in vacuo to approximately ¼ volume, diluted with water and extracted several times with CHCl₃. The organic layers were washed with water, then with saturated NaCl and finally dried over MgSO₄ and concentrated to provide 2-(3,4-dichlorophenoxy)-5-fluoro-benzoic acid as a tan solid, 4.19 g. ¹H-nmr (CDCl₃, 400 MHz, δ): 7.78 (dd, 1H), 7.38 (d, 1H), 7.27 (m, 1H), 6.98 (m, 2H), 6.81 (dd, 1H).

B. 2-(3,4-DICHLOROPHENOXY)-5-FLUORO-BENZOIC ACID ETHYL ESTER

The preceding mixture in 75 mL of ethanol was treated with 1 mL of H₂SO₄ and heated to reflux for 7 hr, then stirred overnight at room temperature. The solvent was removed in vacuo and the residue was dissolved in EtOAc, was with water, saturated aqueous Na₂CO₃ and then water. After drying with MgSO₄, the solvent was evaporated to give a brown oil, 2.1 g. ¹H-nmr (CDCl₃, 400 MHz, δ): 7.64 (dd, 1H), 7.31 (d, 1H), 7.22 (m, 1H), 7.03 (dd, 1H), 6.92 (s, 1H), 6.72 (dd, 1H), 4.24 (q, 2H), 1.19 (t, 3H).

Under N₂, a solution of 12.2 mL (12.2 mmol) of 1.0M lithium bis-(trimethylsilyl)amide in THF (Aldrich Chem. Co.) was stirred and cooled to -30 °C while 781 μL (0.812 g, 7.31 mmol) of 1-vinyl-2-pyrrolidinone (Aldrich Chem. Co.) was added via syringe. After stirring at this temperature for 1 hr, the preceding ethyl ester in 20 mL of THF was added to produce a nearly black solution. This mixture was stirred at room temperature for 72 hr, then diluted with water and EtOAc. The aqueous layer was further extracted with EtOAc and the organic layers were combined, washed with water and saturated NaCl, dried over MgSO₄ and concentrated in vacuo to a tan foam, 2.18 g. This material was used without purification in the following step.

The preceding material was combined with 40 mL of 6N HCl and refluxed for 30 min., producing a gummy precipitate. 1,4-Dioxane (30 mL) was added and the resulting solution was refluxed an additional 30 hr to give a black solution. The reaction was made basic with saturated aqueous K_2CO_3 and extracted three times with EtOAc. The combined organic extracts were washed with water and saturated NaCl, dried with $MgSO_4$ and treated with activated charcoal (Darco G60). Filtration through d.e., washing the pad well with EtOAc and concentration of the filtrate provided a tan oil, 0.803 g, used without further purification in the next step. Mass spectrum (m/e): 325, 323 (m^+).

To 0.300 g (0.93 mmol) of the preceding material in 15 mL of absolute ethanol was added 70 mg (1.86 mmol) of sodium borohydride at room temperature under an N₂ atmosphere (Caution: foaming). After 24 hr, a tlc (CHCl₃:CH₃OH, 95:5) showed a new polar product had
35 formed. Water was added to quench residual sodium borohydride and the mixture was evaporated to a clear oil which was redissolved in EtOAc and washed with water. The organic

layer was further washed with water, saturated NaCl and then dried with MgSO₄. Removal of the solvent in vacuo gave a clear film, 0.202g.

Mass Spectrum (m/e): 328, 326 (m⁺).

¹H-nmr (CDCl₃, 400 MHz, δ): 7.35 (dd, 1H), 7.32 (d, 1H), 6.96 (d, 1H), 6.87 (m, 2H), 6.73 (dd, 1H), 4.26 (t, 1H), 3.11 (m, 1H), 2.97 (m, 1H), 1.78 (m, 2H), 1.52 (m, 1H).

The free base was dissolved in 5 mL Et₂O and treated with 622 mL of 1.0 M HCl in Et₂O, then stirred for 1.5 hr to produce the title compound hydrochloride salt as pale yellow solids, 165 mg, m.p. 171-173 °C

Mass Spectrum (m/e): 328, 326 (m⁺).

Elemental analysis calculated for C₁₆H₁₄Cl₂FNO•HCl: C, 52.99; H, 4.17; N, 3.86. Found: C, 53.23; H, 4.25; N, 3.89.

EXAMPLE 56

[2-(3,4-DICHLOROPHENOXY)-5-FLUOROBENZYL]-METHYLAMINE.

Under N₂, a mixture of 0.313g (1.0 mmol) of 2-(3,4-dichlorophenoxy)-5-fluoro-N-methylbenzamide in 5.0 mL of anhydrous tetrahydrofuran (THF) was treated via syringe with 4.0 mL (4.0 mmol) of 1.0 M BH₃ in THF (Aldrich Chem. Co.) and the mixture was heated to reflux for a total of 48 hr. The reaction was quenched by the addition of 25 mL of 6N HCl and heating to reflux until the free amine could be detected by tlc (CHCl₃:CH₃OH:TEA, (95:5:1)). The cooled mixture was then diluted with water, basified with K₂CO₃ and extracted with EtOAc. The combined organic layers were washed with water and saturated NaCl, then dried with MgSO₄, filtered and concentrated in vacuo to the free base as a light brown oil, 0.164 g, 54%). This compound was converted to the hydrochloride salt as described previously, m.p. 200-202 °.

EXAMPLE 57

2-(3,4-DICHLOROPHENOXY)-5-FLUORO-BENZYLAMINE

A flame-dried flask containing N₂ inlet and magnetic stirrer was charged with 3.0 mL of 2.0 M borane methyl sulfide complex in THF (6.0 mmol, Aldrich Chem. Co.), followed by an additional 10 mL of anhydrous THF at room temperature. While stirring, 0.562 g (2.0 mmol) of 2-(3,4-dichlorophenoxy)-5-fluoro-benzonitrile (title compound of Preparation 37) was added slowly via syringe, causing some mild foaming. After the addition, the reaction was heated at reflux for a total of 3 hr, and a tlc (CHCl₃:CH₃OH:conc. NH₄OH, 95:5:1) showed no starting material remaining. With ice bath cooling, the reaction was quenched using 10 mL of 6N HCl, heated to reflux an additional 1 hr to break the boron complex and slowly made basic with saturated aqueous Na₂CO₃. The mixture was diluted with water and EtOAc, the organic layer was combined with a second EtOAc extraction of the aqueous layer and then washed with water and saturated aqueous NaCl. After drying over MgSO₄, the solvent was removed in vacuo to give a yellow tan oil, 0.676 g. The oil was partitioned between Et₂O and 6N HCl, the

Et₂O layer was reextracted with 6N HCl and the aqueous layers were combined, made basic with aqueous Na₂CO₃ and reextracted with Et₂O. These latter organic extracts were dried (MgSO₄) and concentrated in vacuo to a tan oil, 0.538 g.

Mass spectrum (APCI, m/z): 286 (m⁺), 288.

- 5 The oil was dissolved in Et₂O and treated with 2.0 mL of 1.0 M HCl in Et₂O. The resulting solids were stirred at room temperature for 2 hr, filtered, washed with Et₂O and dried under vacuum to give the hydrochloride salt of the title product, 0.434 g, m.p. 190-194 °C.

Elemental Analysis calculated for C₁₃H₁₀Cl₂FNO•HCl: C, 48.40, H, 3.43, N, 4.34. Found: C, 48.22, H, 3.80, N, 4.28.

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